

# Starting Soon

## Practical Approaches for PFAS Fate & Transport Evaluation Training

ITRC Resources:

<https://pfas-1.itrcweb.org/>

CLU-IN Training Page (slides available):

<https://www.clu-in.org/conf/itrc/PFAS-fate-transport/>



# Housekeeping

This event is being recorded; Event will be available On Demand after the event at the main training page

<https://www.clu-in.org/conf/itrc/PFAS-fate-transport/>

If you have technical difficulties, please use the Q&A Pod to request technical support

Need confirmation of your participation today?

Fill out the online feedback form and check box for confirmation email and certificate

# ITRC – Shaping the Future

- Host Organization   
ECOS
- Network - All 50 states, PR, DC

- Federal Partners     
DOE DOD EPA

- ITRC Industry Affiliates Program 

- Academia

- Community Stakeholders

## Disclaimer

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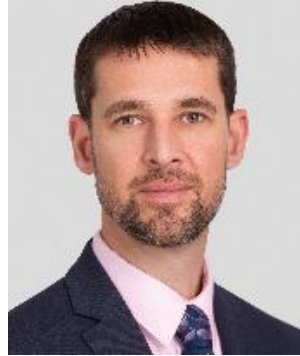
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# Meet The Trainers



**Sarah LaRoe**  
Anchor QEA  
[slaroe@anchorqea.com](mailto:slaroe@anchorqea.com)



**Matt Machusick**  
Leidos  
[machusickm@leidos.com](mailto:machusickm@leidos.com)



**Li Wang**  
CA DTSC  
[Li.Wang@dtsc.ca.gov](mailto:Li.Wang@dtsc.ca.gov)



**Dina Drennan**  
BEM Systems  
[ddrennan@bemsys.com](mailto:ddrennan@bemsys.com)



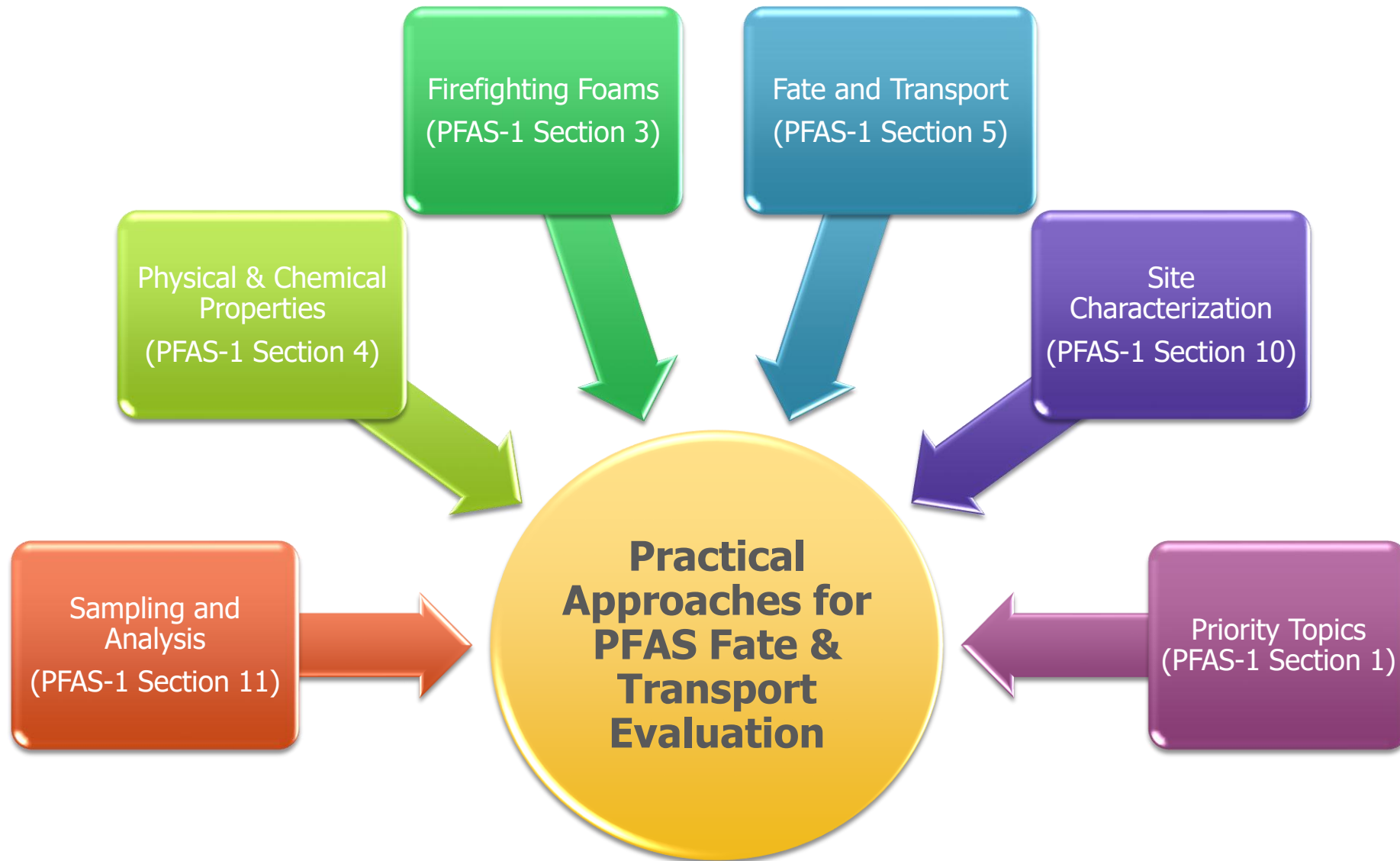
**Alex MacDonald**  
Emeritus  
[alexmacd836@gmail.com](mailto:alexmacd836@gmail.com)




# Practical Approaches for PFAS Fate & Transport Evaluation



# Materials Used Today



# Training Scope



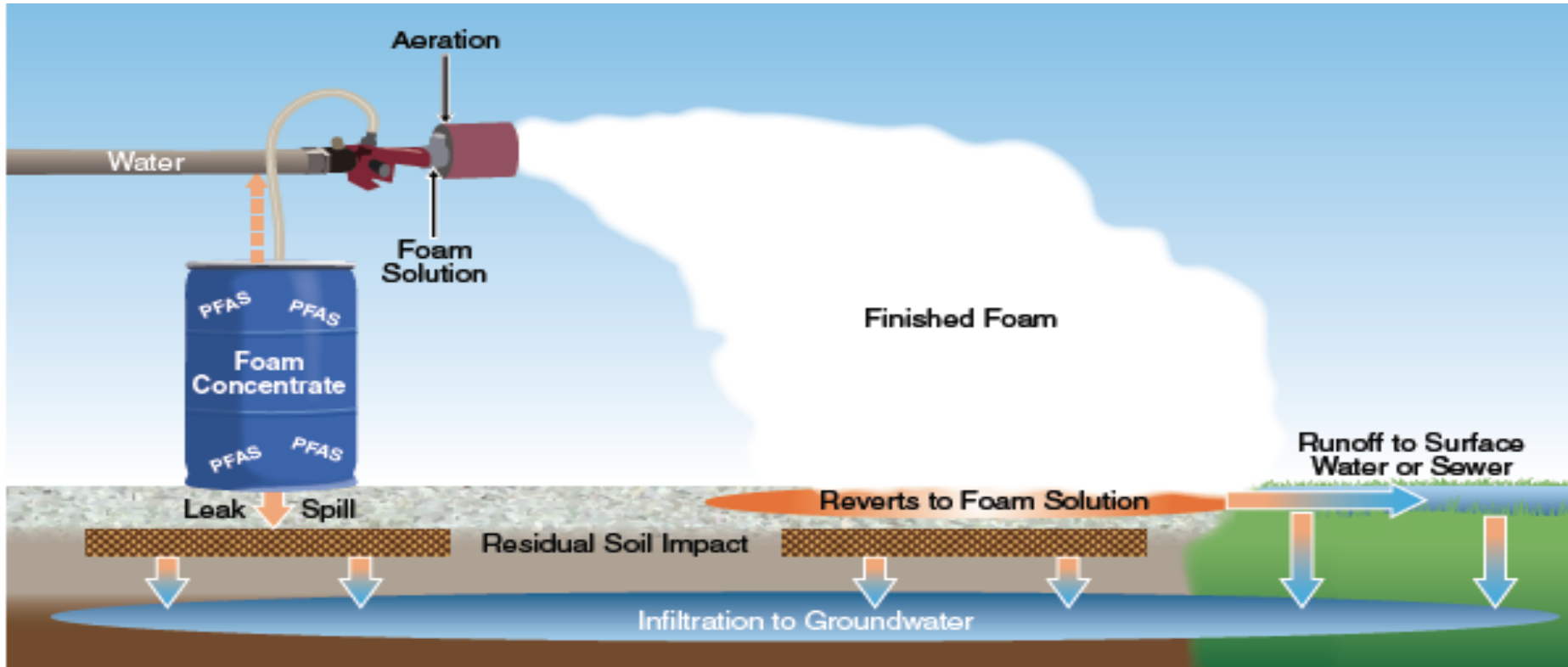
**Practical  
Approaches  
for PFAS Fate  
& Transport  
Evaluation**

Review of topics that apply to a hypothetical aqueous film-forming foam (AFFF) release scenario

- Includes discussion of specific site characteristics and relevant environmental compartments
- Not a comprehensive look at the hypothetical scenario

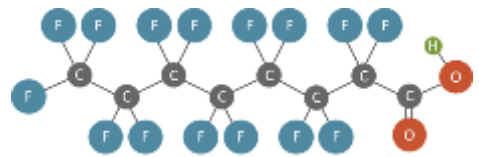


# Hypothetical Source: Class B AFFF

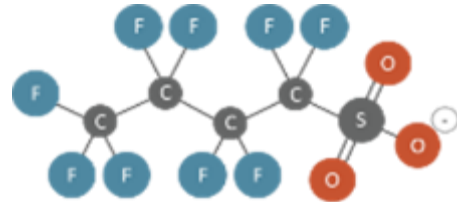


PFAS-1 Figure 3-4 Release of firefighting foam.  
Source: Adapted from figure by J. Hale, Kleinfelder, used with permission.

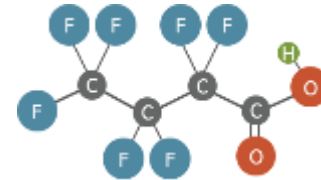
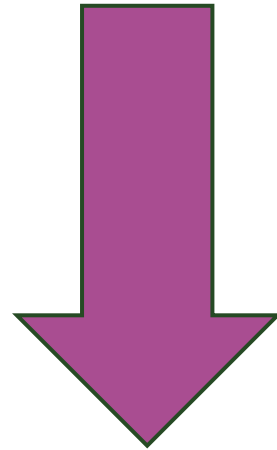
# PFAS Chemistry is Unique



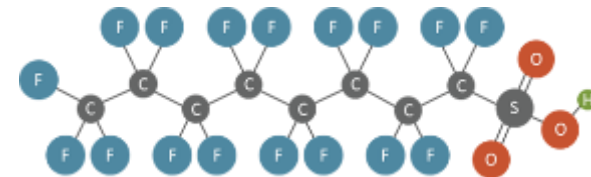
PFOA (perfluorooctanoic acid)



PFBS (Perfluorobutane sulfonic acid)



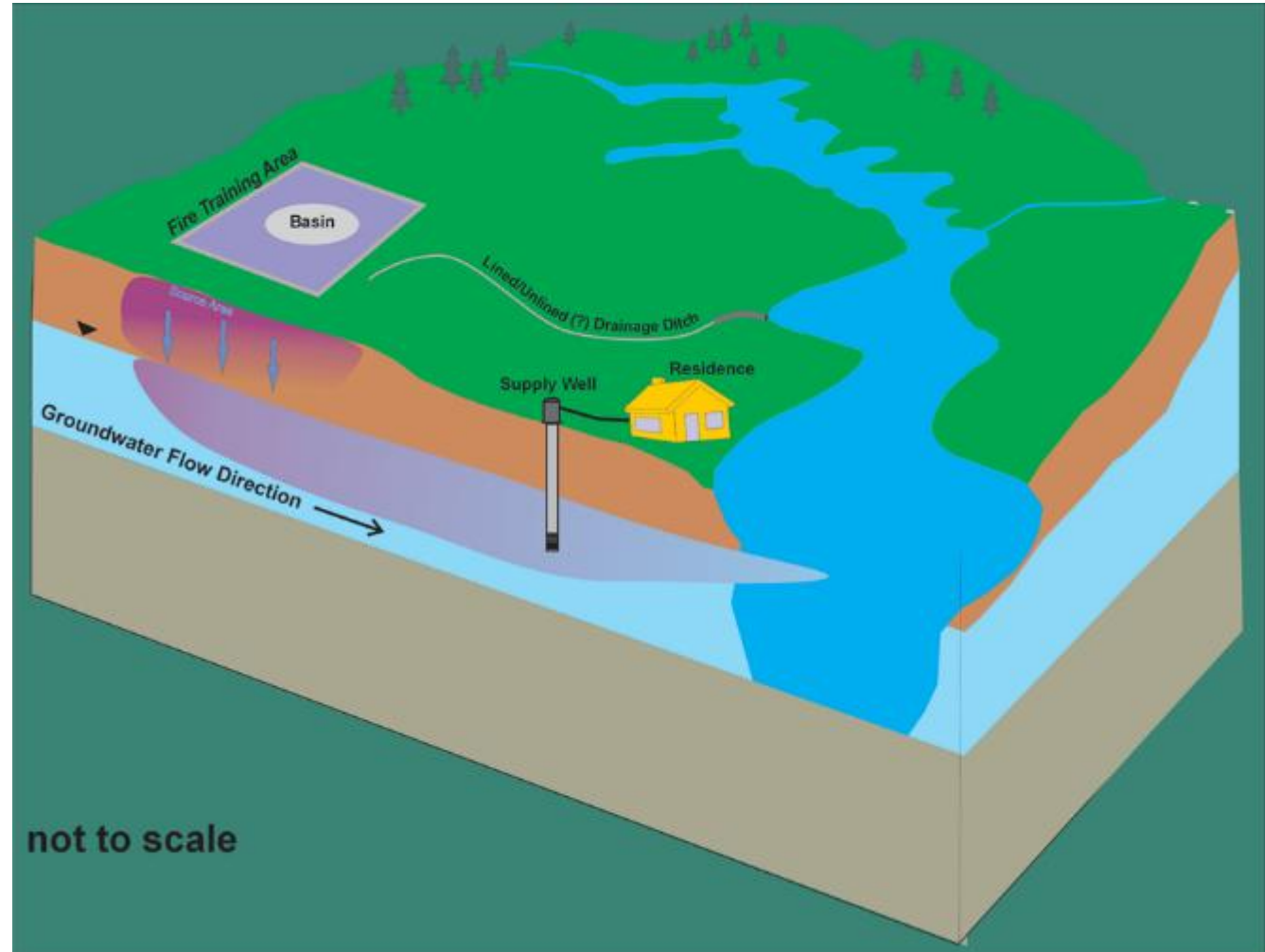
PFBA (Perfluorobutanoic acid)



Perfluorooctane sulfonic acid (PFOS)

*How does that Translate to Site Dynamics?*

# The Site – AFFF Firefighting Training Facility



Source: Alex MacDonald. Used with Permission.

# Case Study – AFFF Firefighting Training Facility

## Facility Attributes

- Hypothetical unlined 8-acre facility operated from around 1970 to 1999
- Used to train local firefighters in using AFFF
- Main feature - a concrete basin to float burning fuel
- Residential community downgradient uses groundwater for domestic and agricultural supply
- Groundwater discharges to surface water

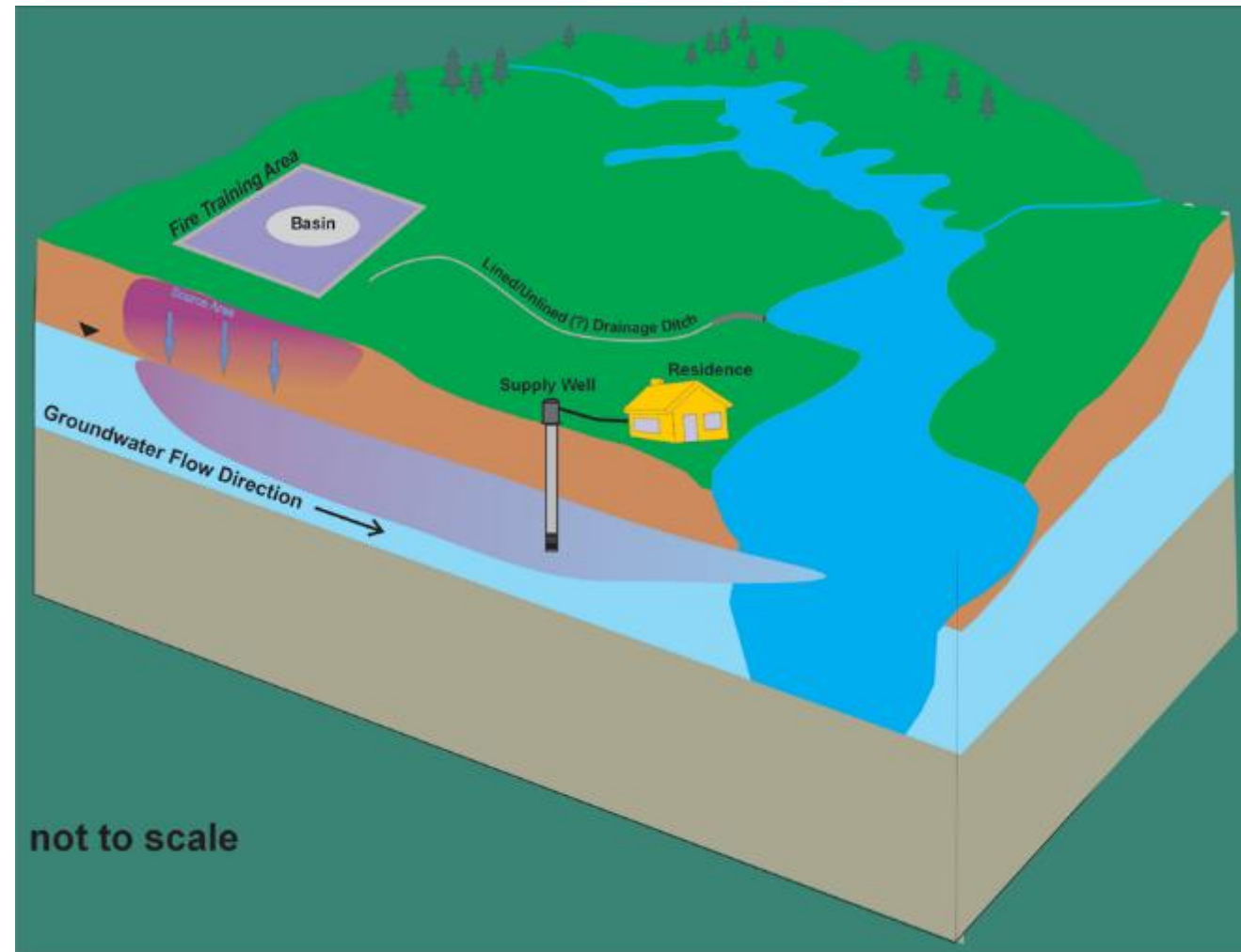


Source: Alex MacDonald. Used with Permission.

# Case Study – AFFF Firefighting Training Facility

## Facility and Surrounding Area 3-D View

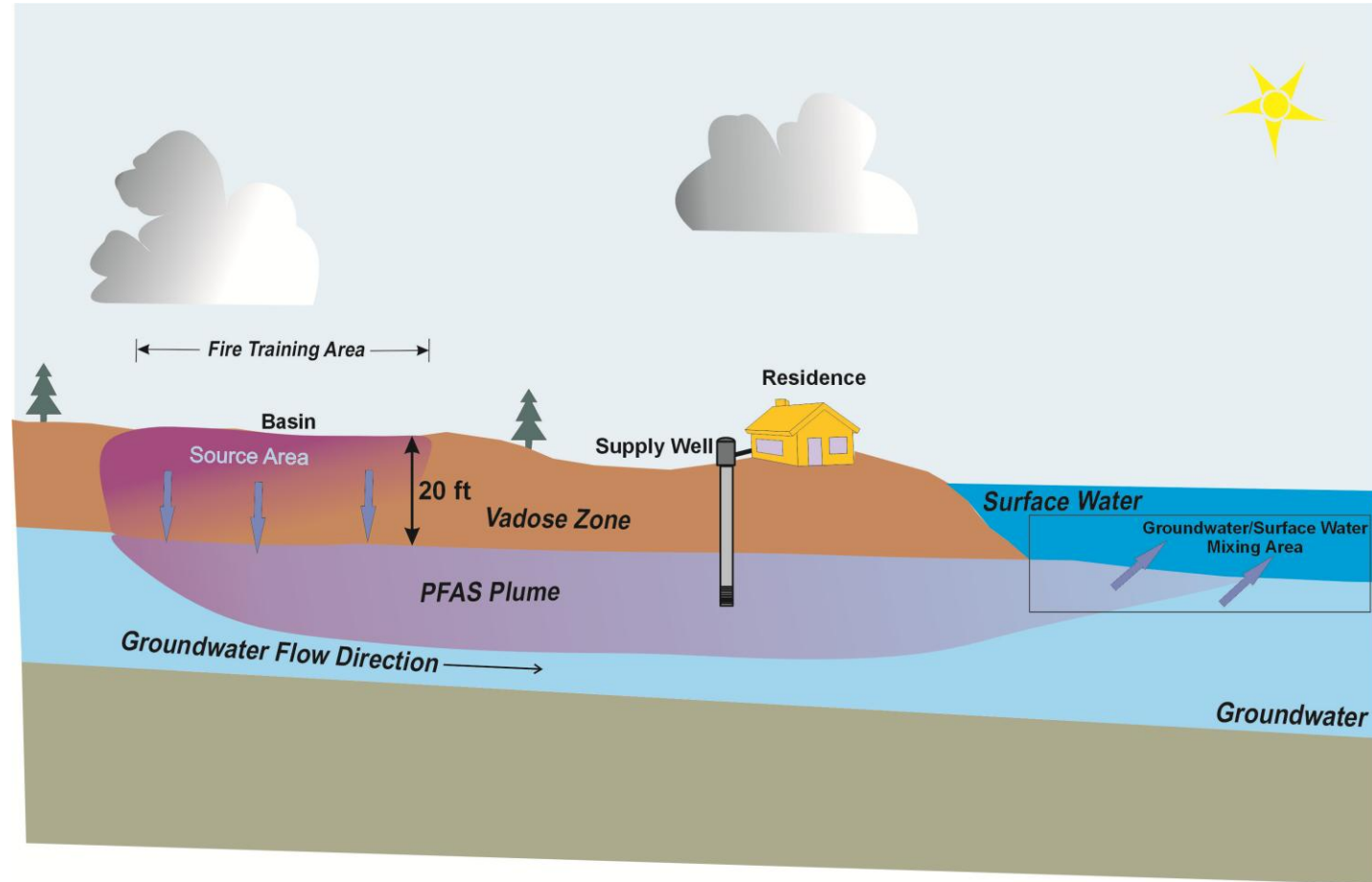
- Light purple area – former training facility with concrete apron
- Community is represented by the single house with supply well
- Interim treatment has been installed on water supply wells while the investigation is implemented
- Potential PFAS discharge to surface water during storm events
- Potential human exposures to PFAS



# Case Study – AFFF Firefighting Training Facility

## Cross-Sectional View

- 20-foot vadose zone
- Fluctuating water table – 5ft
- Subsurface consists of heterogeneous unconsolidated sediments
- Saturated zone approximately 30 ft thick, underlain by impermeable confining layer
- PFAS in groundwater discharges to surface water
- Average annual recharge 15 inches/year
- Nearest well 300 feet downgradient

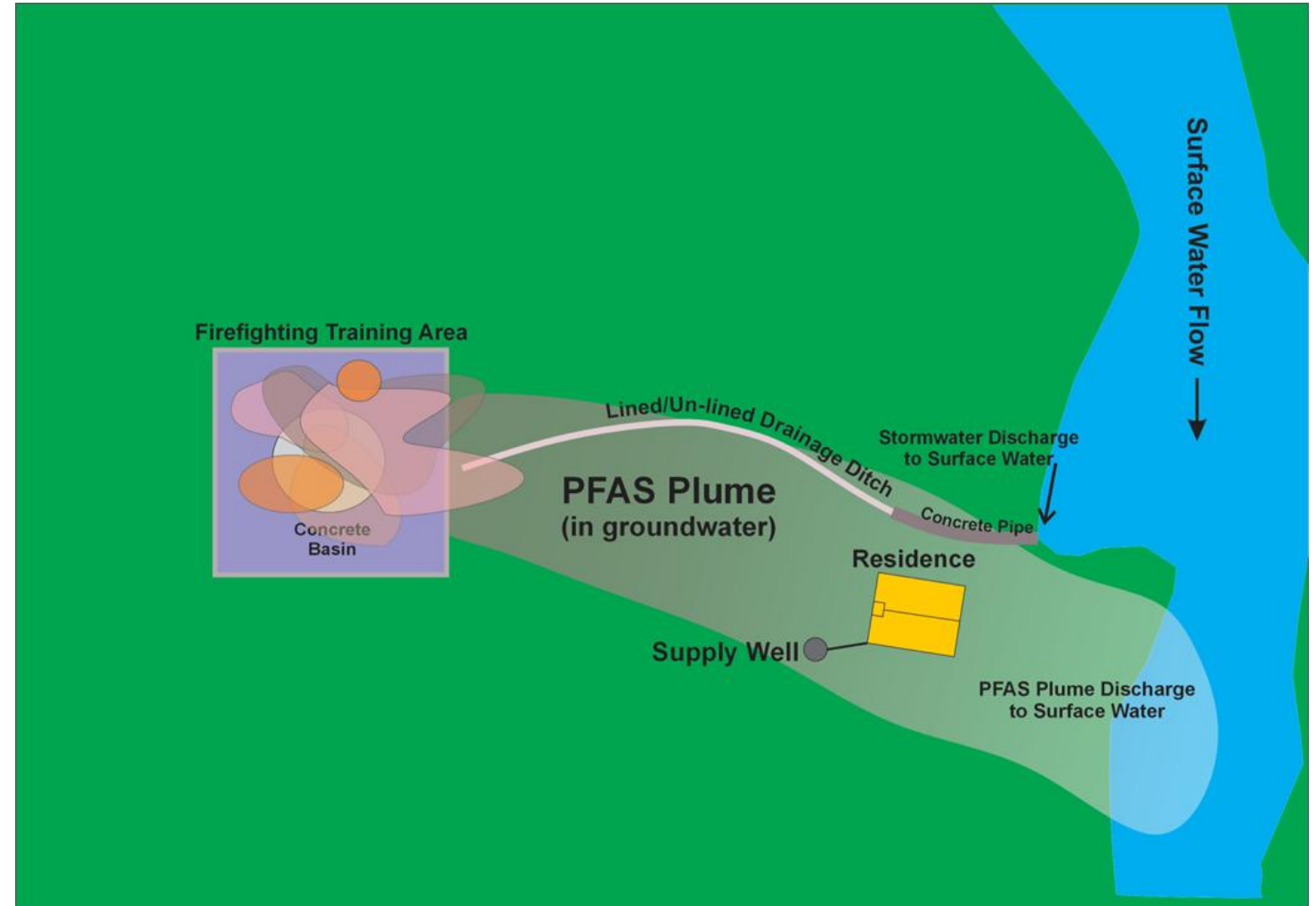


Source: Alex MacDonald. Used with Permission.

# Case Study – AFFF Firefighting Training Facility

## Overhead View

- Note – different shapes in the training area
- Fuels used for generating fire are another pollutant for investigation
- Stormwater from the training area travels a lined/unlined drainage ditch to surface water



# Case Study – Goals of the Investigation

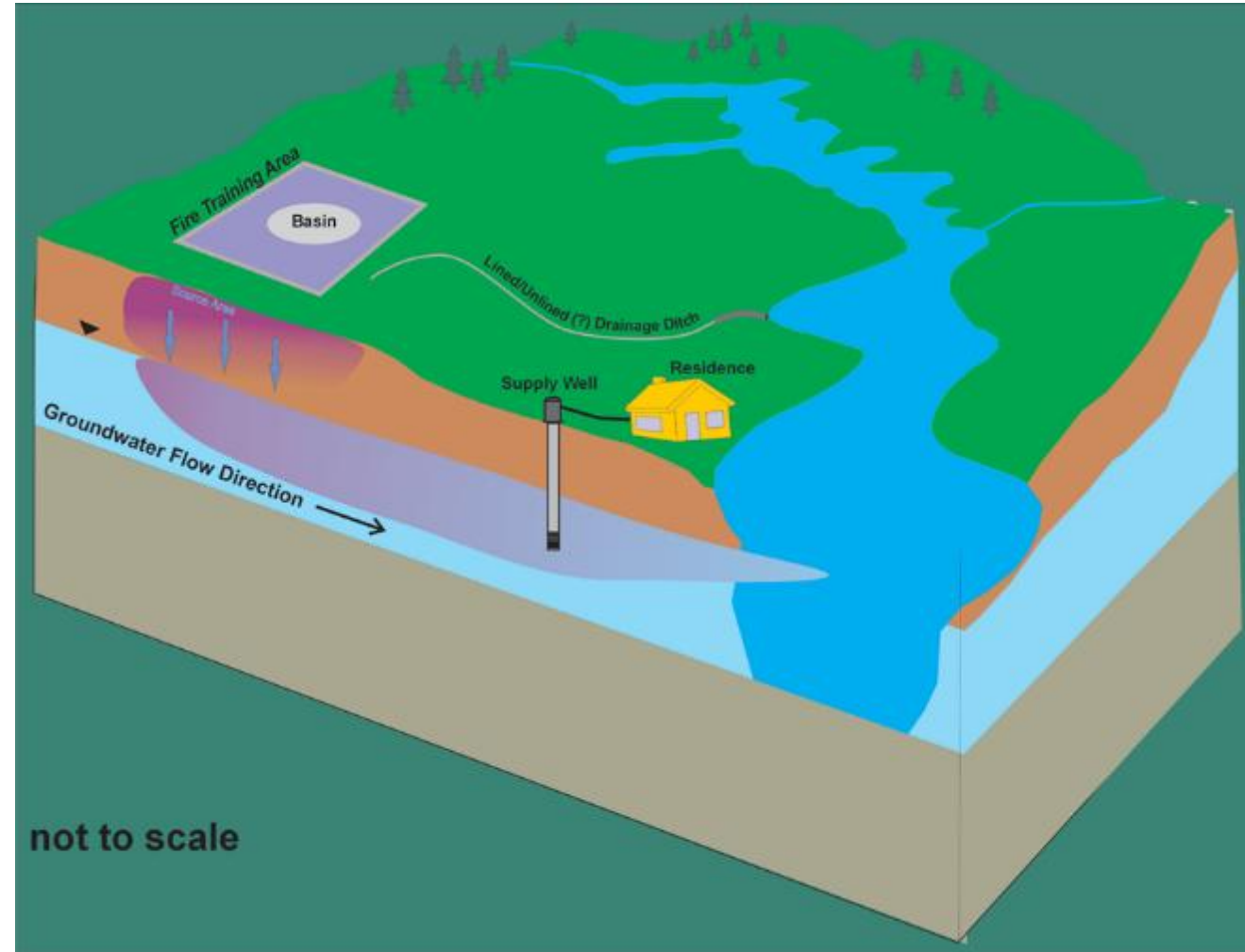
Gather available information

Develop conceptual site model

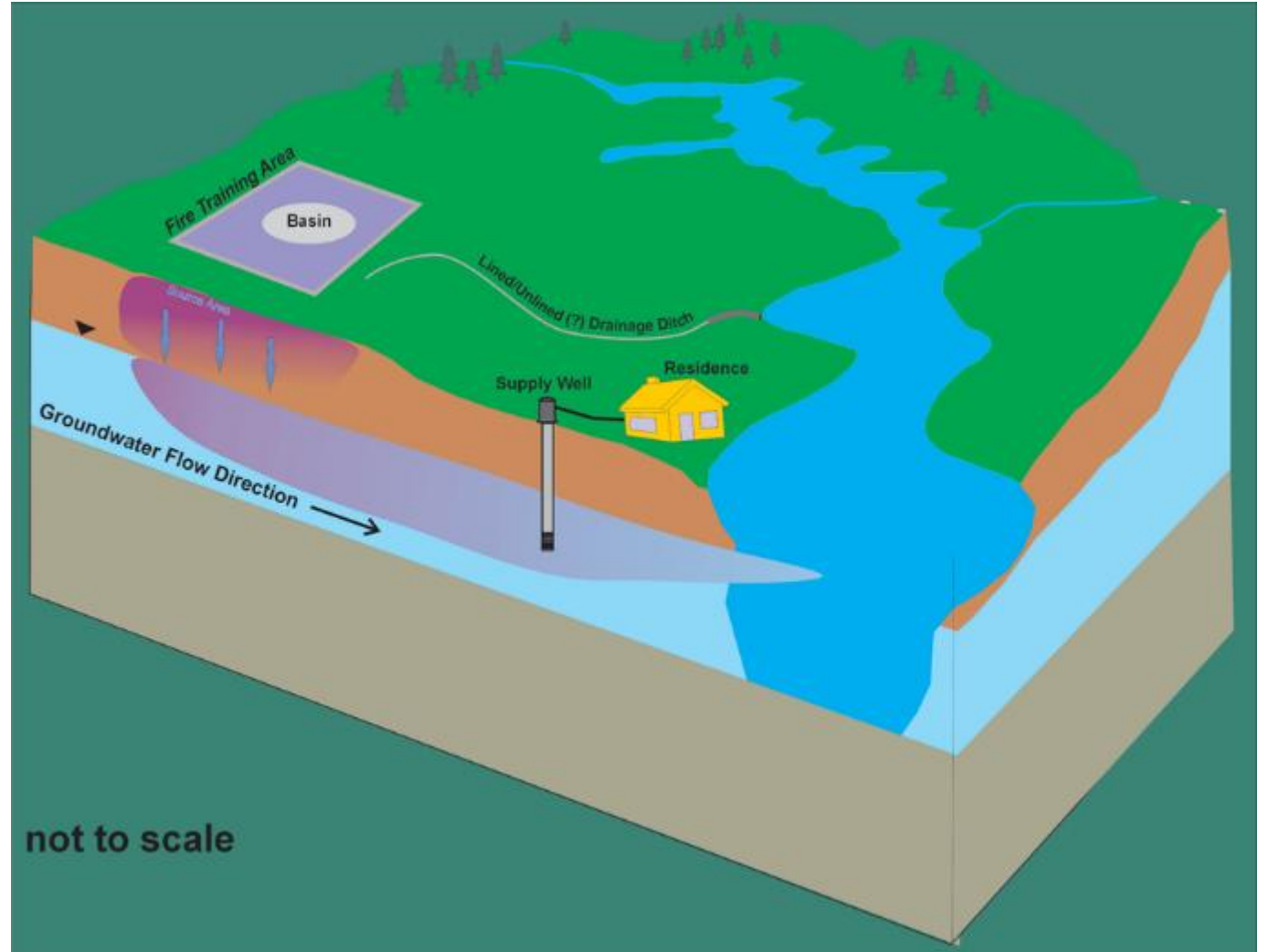
Identify human and ecological receptors

Produce a Sampling and Analysis Plan to:

- Determine extent of COCs
- Evaluate impacts to groundwater and surface water
- Locate and measure residual COC concentrations/mass
- Determine if sources are controlled
- Evaluate if regulatory values are exceeded
- Assess human and ecological risks



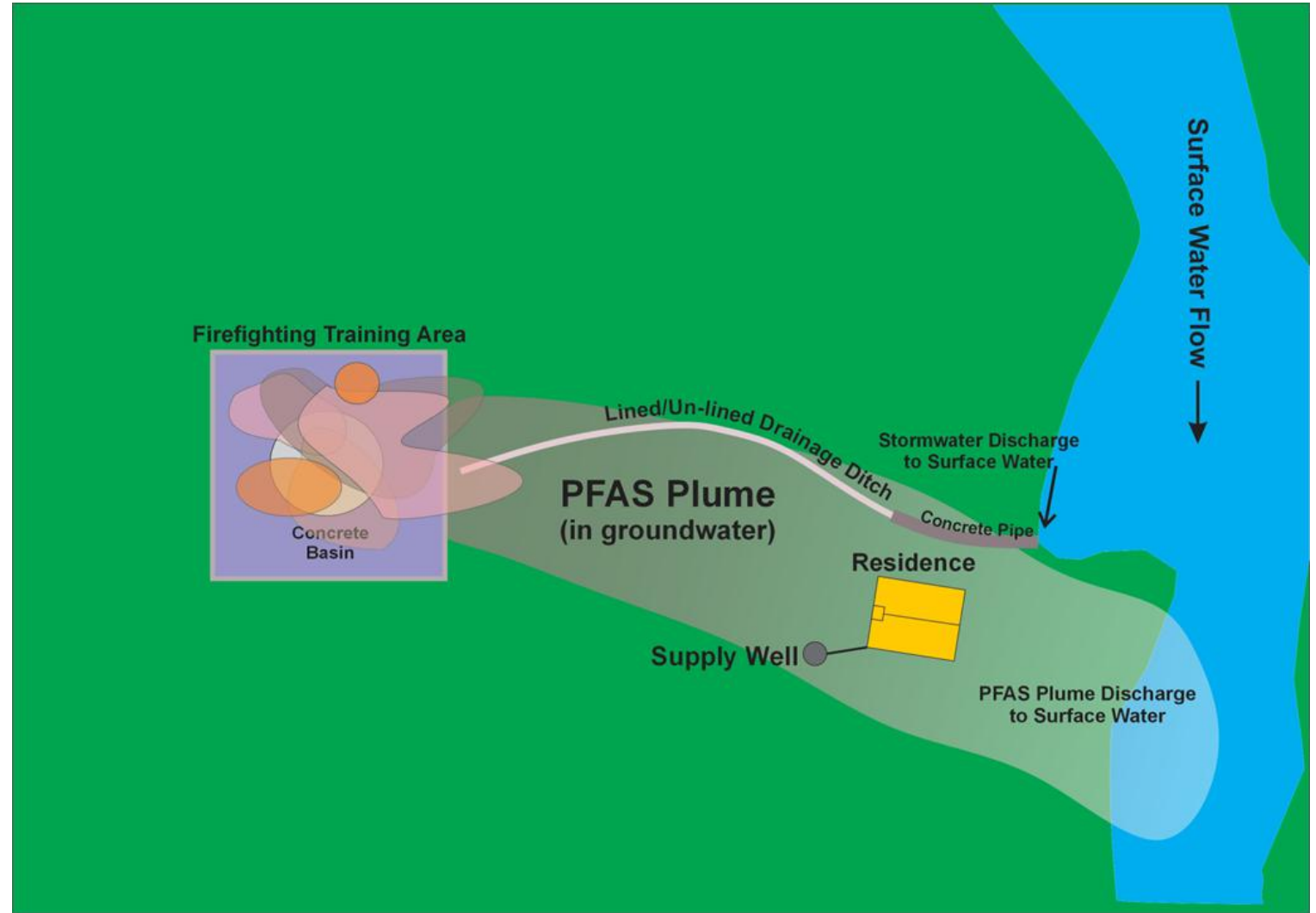
# Source Zone Characterization



# Source Zone Characterization

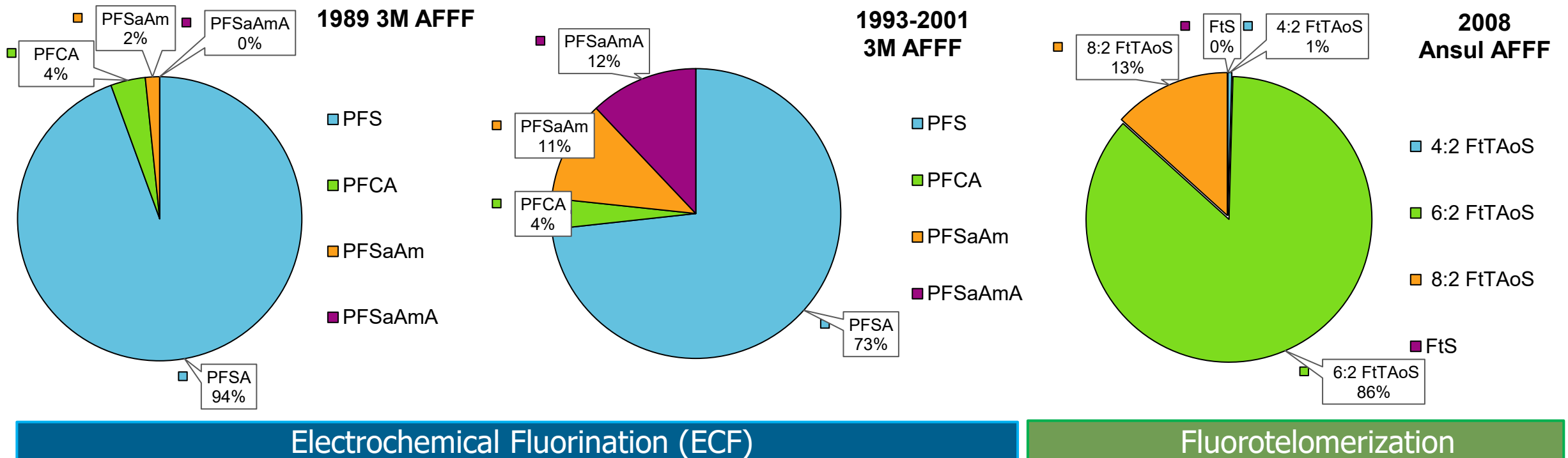
## Source area

- Area of contaminated media that continues to contribute mass to the PFAS plume
- Area results from a number of cumulative releases over time
- Defined on horizontal and vertical planes
- Likely includes concrete pad and surrounding surface soil



# Class B AFFF Contains Highly Diverse Mixtures

## AFFF product chemistry has changed over time



Electrochemical Fluorination (ECF)

Fluorotelomerization

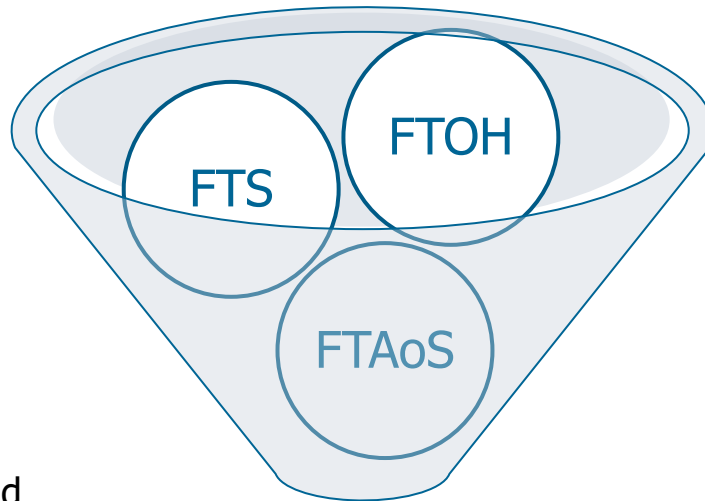
PFAS-1 Table 3-1, Types of Foam and Composition.

Figures used with permission from J. Field, Oregon State.

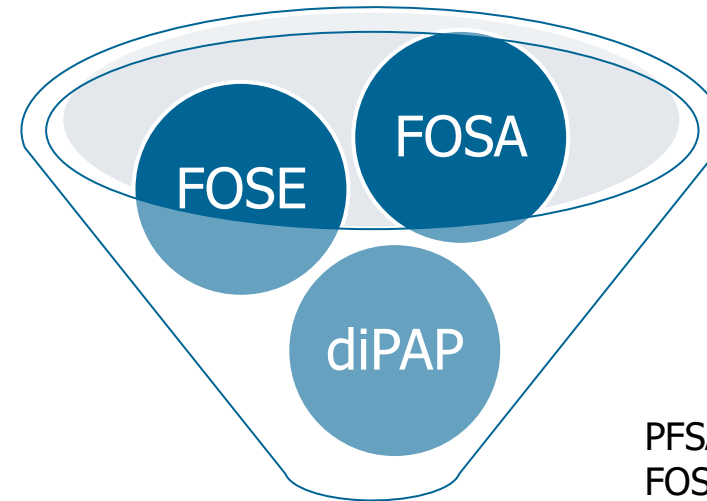
# Source Zone Characterization

## Understand AFFF source material

- Precursor composition determines fate and transport potential and final composition



↓  
**PFCAs**



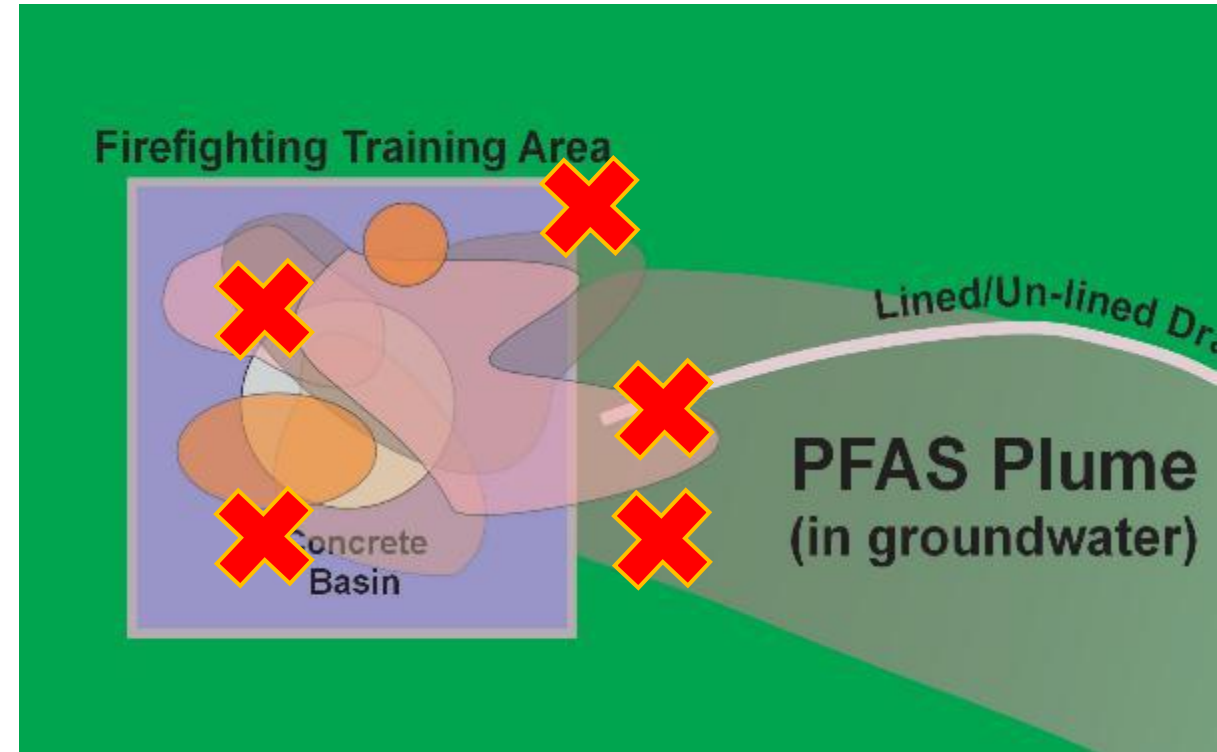
↓  
**PFSAs**

PFSA: Perfluorosulfonic Acid  
FOSE: Perfluorooctane sulfonamidoethanol  
FOSA: Perfluorooctane sulfonamide  
diPAP: Polyfluoroalkyl phosphate diester

PFCA: Perfluorocarboxylic Acid  
FTS: Fluorotelomer Sulfonic Acid  
FTOH: Fluorotelomer Alcohol  
FTAoS: Fluorotelomer thioether amido sulfonate

# Source Zone Characterization – Sampling Plan

- Understand extent by sample placement
  - Concrete pad and apron
  - Vertical soil profiles to understand extent and distribution of infiltration
  - Sediment and concrete in drainage channels
- Understand composition through choice of analytical method
  - Different analytical methods can determine precursor composition
    - EPA Method 1633A
    - Non-target Analysis
    - Total Oxidizable Precursor (TOP) Assay/Total Organic Fluorine (TOF)
- Results inform potential source control design



 Sample Location

# Knowledge Check!

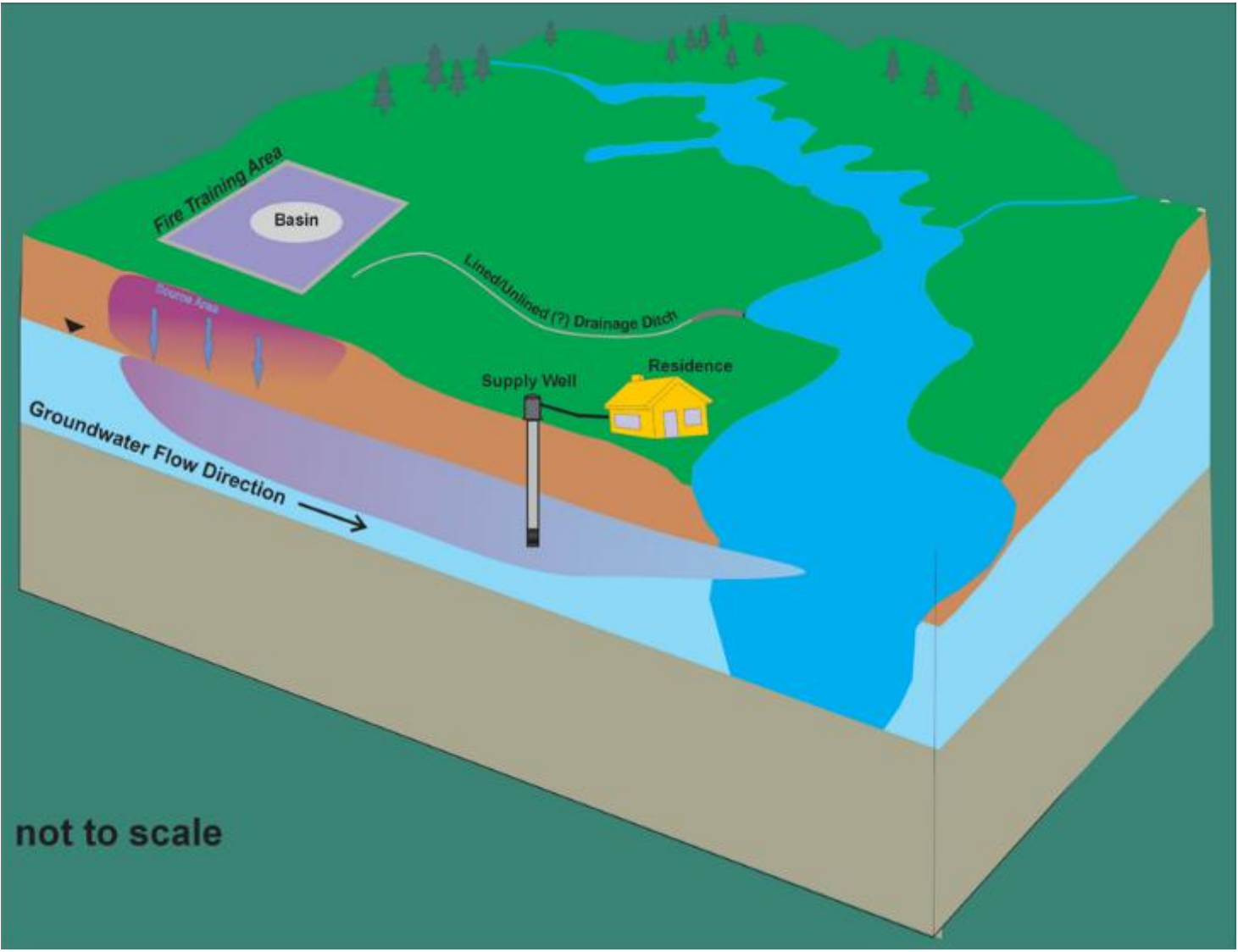
Check  
In!

AFFF composition can vary due to:

- A. Manufacturing process that produced the PFAS
- B. The AFFF supplier that produced the formulation
- C. When the AFFF supplier produced the formulation
- D. All of the above



# Vadose Zone

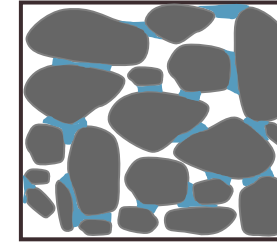


# PFAS Behavior in the Vadose Zone

- Surfactant behaviors
  - Air-Water interfacial (AWI) retention
  - Amplified relative to hydrocarbon surfactants
  - Particularly important for longer chain & sulfonates (PFOS vs PFBA)
- Interactions with Soil
  - Organic carbon is only part of the story
  - pH
  - Divalent cations
  - Remedial history
- Source strength considerations
  - Typical Dilution Attenuation Factor (DAF) not representative

## Shallow Source Area

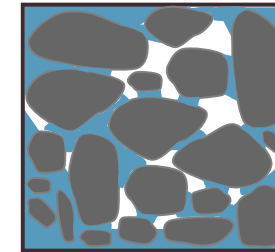
*Long chain PFAS partition to interfaces & Significant retention*



- Underlies source zone
- Persistent case in arid environments

## Recharge

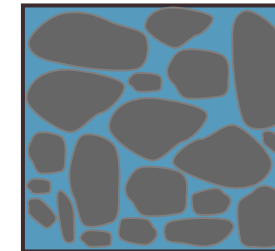
*Partitioning into aqueous phase & Competitive sorption*



- Transient condition
- Frequency dependent on climate
- Persistent case in humid environments

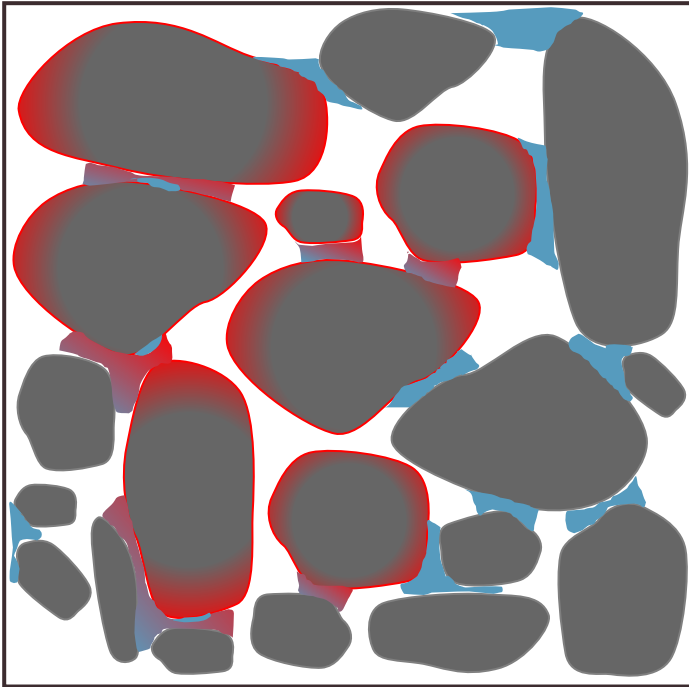
## Saturation AWI~0

*AWI collapse, Mass release, & Colloid transport*

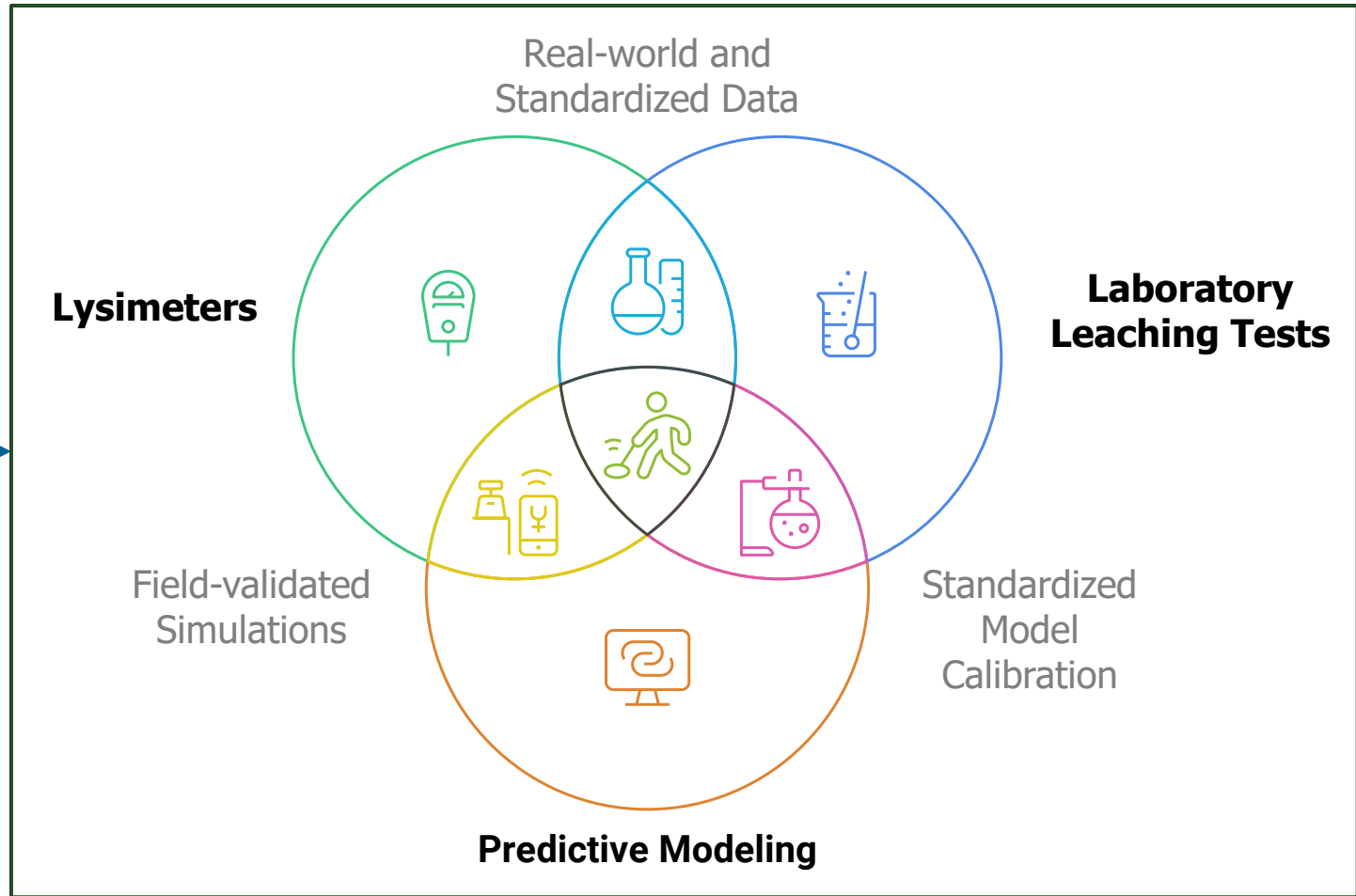


- Transient condition may be seasonal and brief
- Hysteresis (matrix diffusion through newly connected pores)

# Sampling Plan → Source Strength Evaluation



What soil concentration is expected to yield a leachate concentration protective of groundwater?



# Leaching and Lysimeter Tests

## Leaching Tests

- Do not account for PFAS retention at the air-water interface
- Do not represent the soil-to-leachate ratio encountered under field conditions
- Can address kinetic desorption (LEAF)
- SPLP derived  $K_d^1$

## Lysimeter Porewater

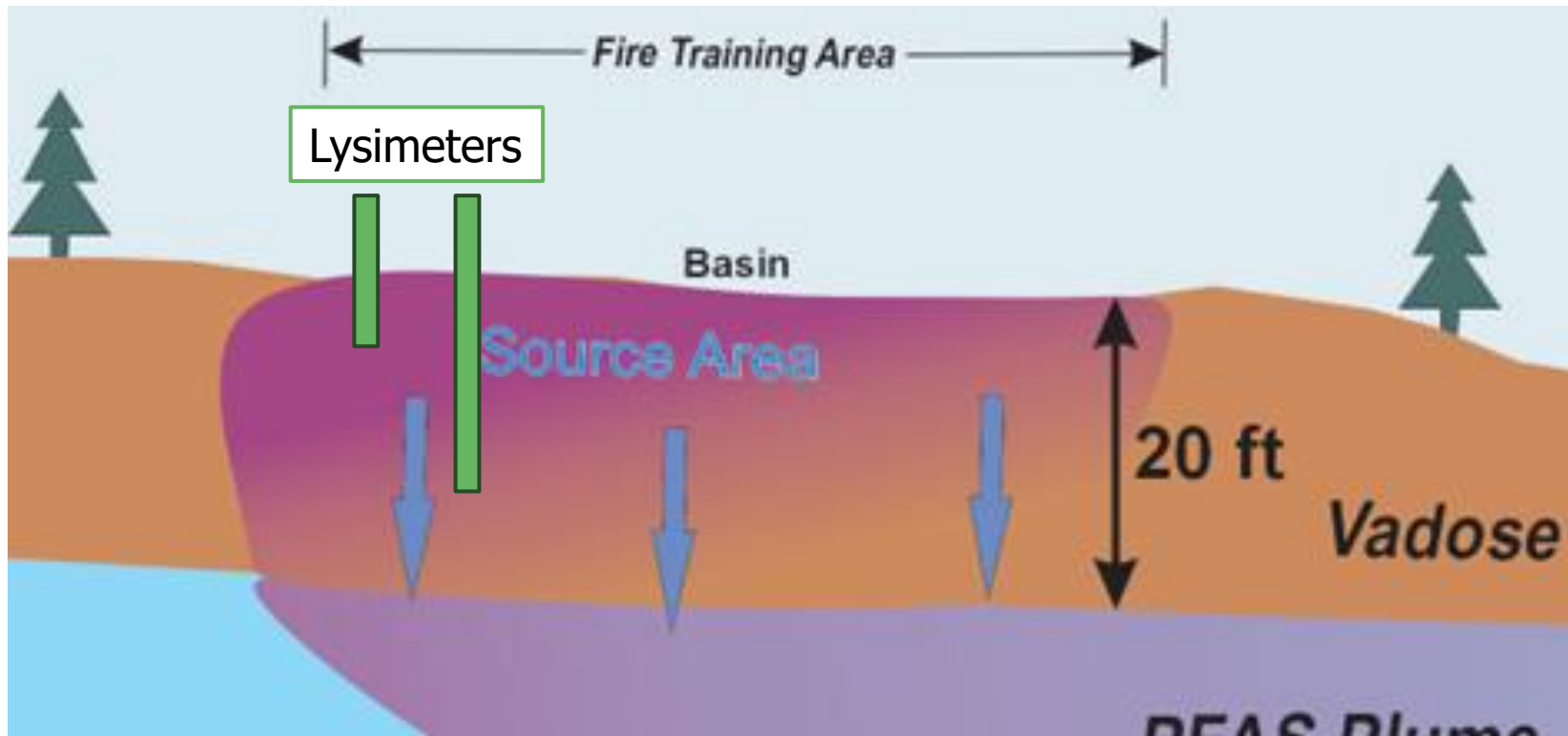
- Represent a single time result → consider this from season to season
  - Relative to soil moisture
- Limitations include:
  - Very shallow groundwater
  - Bedrock
  - Very arid environments
  - Insufficient co-located soil sample data

### BOTTOM LINE

- Soil concentration alone does not account for retentive properties → misleading for source strength
- Conservative DAF likely overestimates true source strength
- Lysimeters can improve interpretation, if feasible

<sup>1</sup> NJDEP. 2025. "PFAS SPLP Calculator." <https://dep.nj.gov/srp/guidance/rs/>  
(See also PFAS-1 Section 1.5.1)

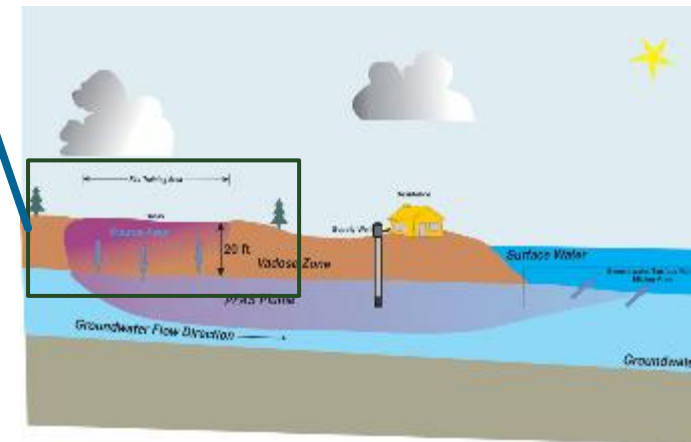
# Vadose Zone Characterization\*



- Lysimeter installation within the source zone.
- Shallow (below the root zone)
- Deep (above the capillary fringe)
- Soil samples coincident with suction cup (splits)
  - Laboratory leaching tests
  - Total PFAS
  - Characterization (model support)

How many lysimeters are needed?

- Range of depths
- Range of permeabilities
- Range of concentrations



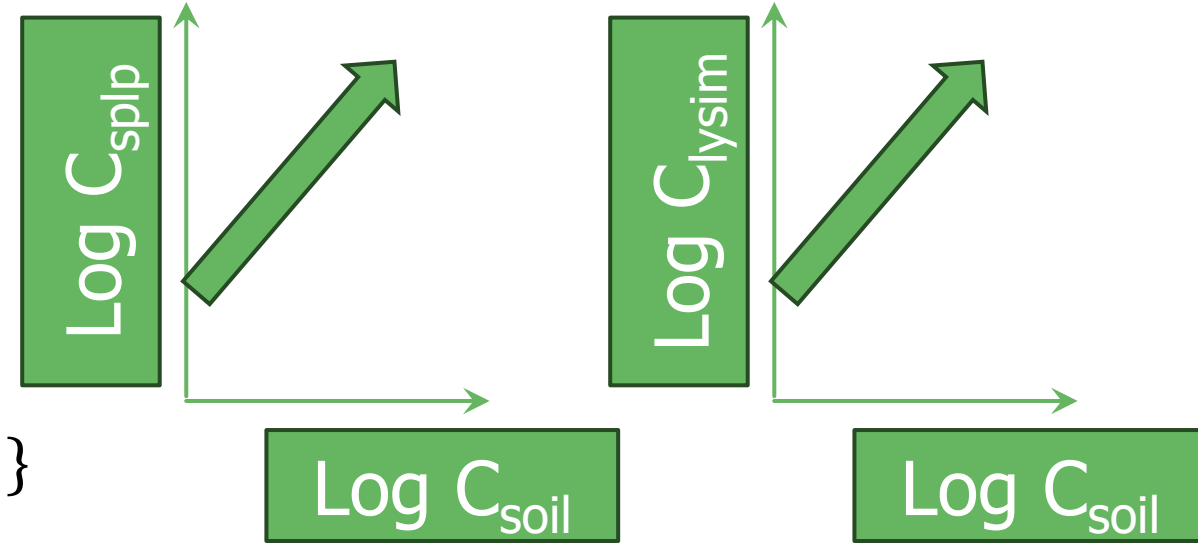
**\*Our CSM is conducive to lysimeter sampling; this may not always be the case!**

# Example site source strength evaluation

Eqn 1:  $PW_{PGW} = DAF * PSL$

Eqn 2:  $* DAF = 1 + \left(\frac{Kid}{IL}\right)$

Eqn 3:  $d = (0.0112 * L^2)^{0.5} + d_a \left\{ 1 - \exp \left[ \frac{(-LI)}{(Kid_a)} \right] \right\}$



## Characterization Data (CSM)

- K = aquifer hydraulic conductivity (meters per year [m/yr])
- i = hydraulic gradient (meters per meter [m/m])
- d = mixing zone depth (meters [m])
- I = infiltration rate (m/yr)
- L = source length parallel to groundwater flow (m)
- d<sub>a</sub> = aquifer thickness (m)

Eqn 4:  $Leachate_{PFAS} = mC_{soil} + b$

Eqn 5:  $SSL = 10 \left( \frac{\log PW_{pgw} - b}{m} \right)$

SSL: Soil Screening Limit; PW: Porewater; PGW: Protective of groundwater; PSL: Project Screening Limit

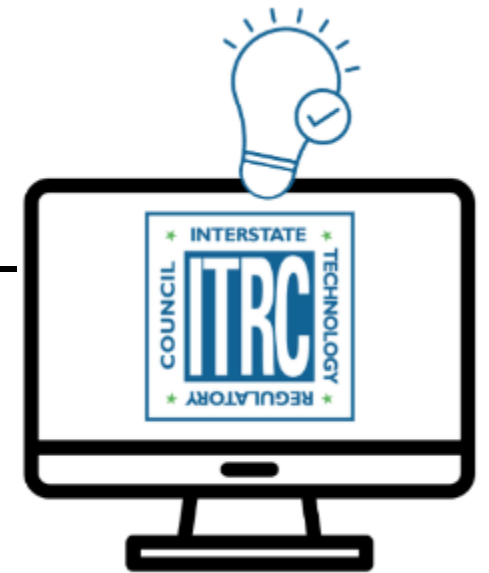
\*Source of the DAF equations: USEPA 1996. Soil Screening Guidance: Users Manual, 2nd Edition. Publication 9355.4-23. July

# Knowledge Check!

Check  
In!

Why would you expect a PFOS SSL derived from SPLP to be higher than an SSL derived from lysimeter PW?

- a) Collapse of the air-water interface under fully saturated conditions during SPLP results in mass release.
- b) Lysimeters do not pull representative samples under certain conditions and SSLs derived from lysimeter PW are always biased low.
- c) A long chain sulfonate like PFOS would not have a higher SPLP-derived SSL. They should be the same.

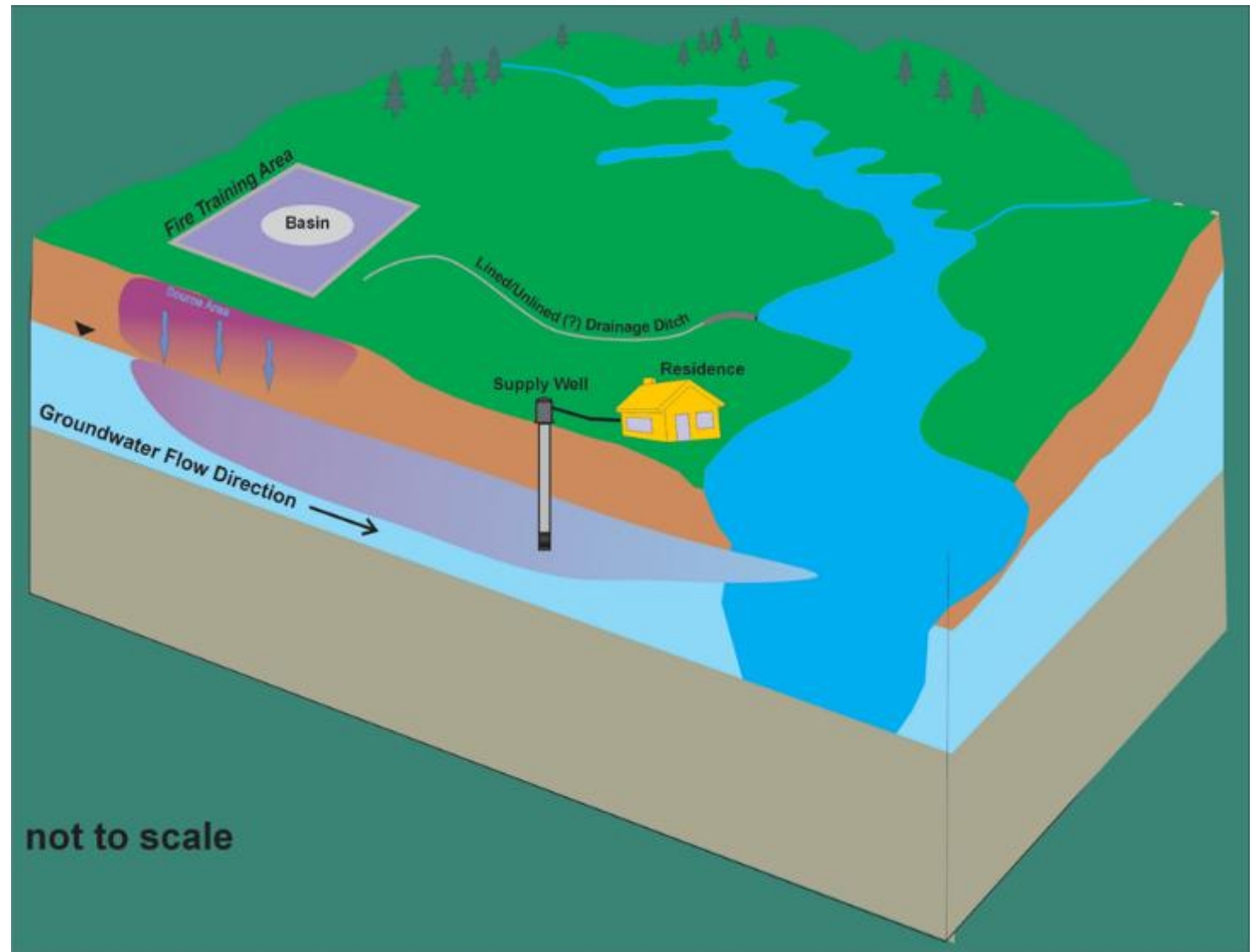


# Question and Answer Break

Please use the Q&A Pod to ask questions.

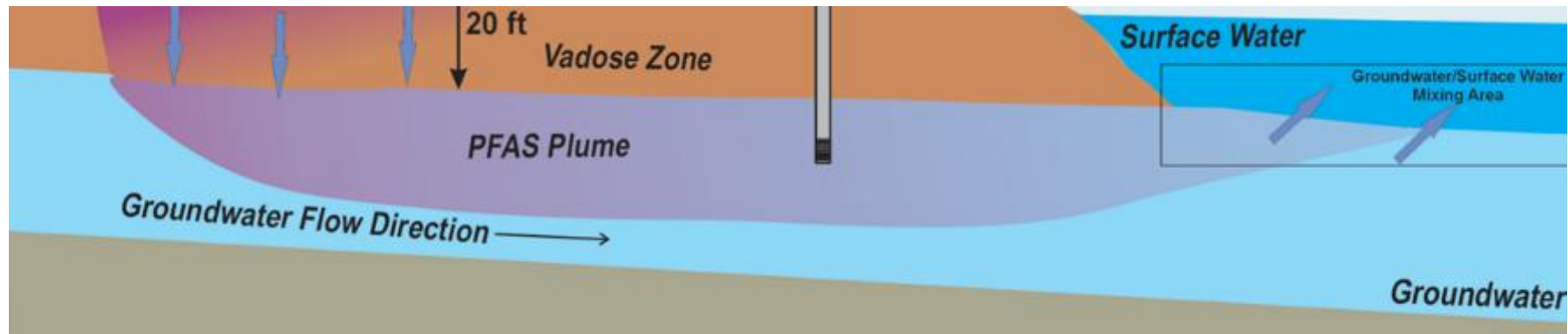
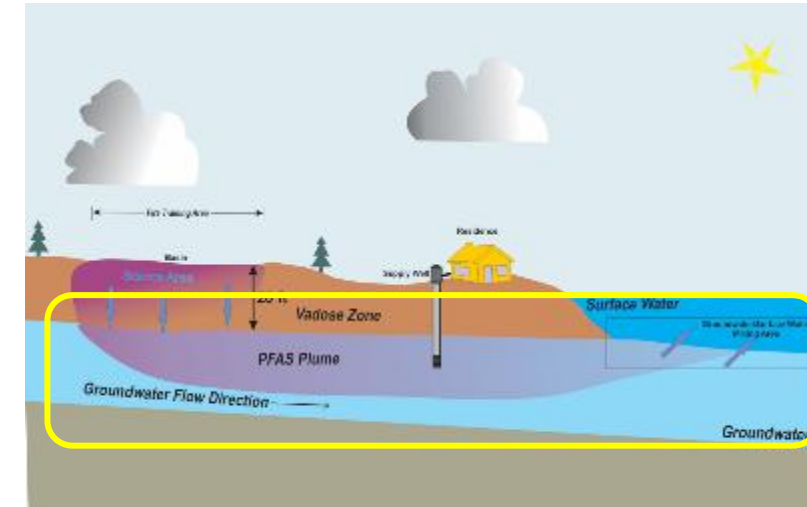


# Saturated Zone

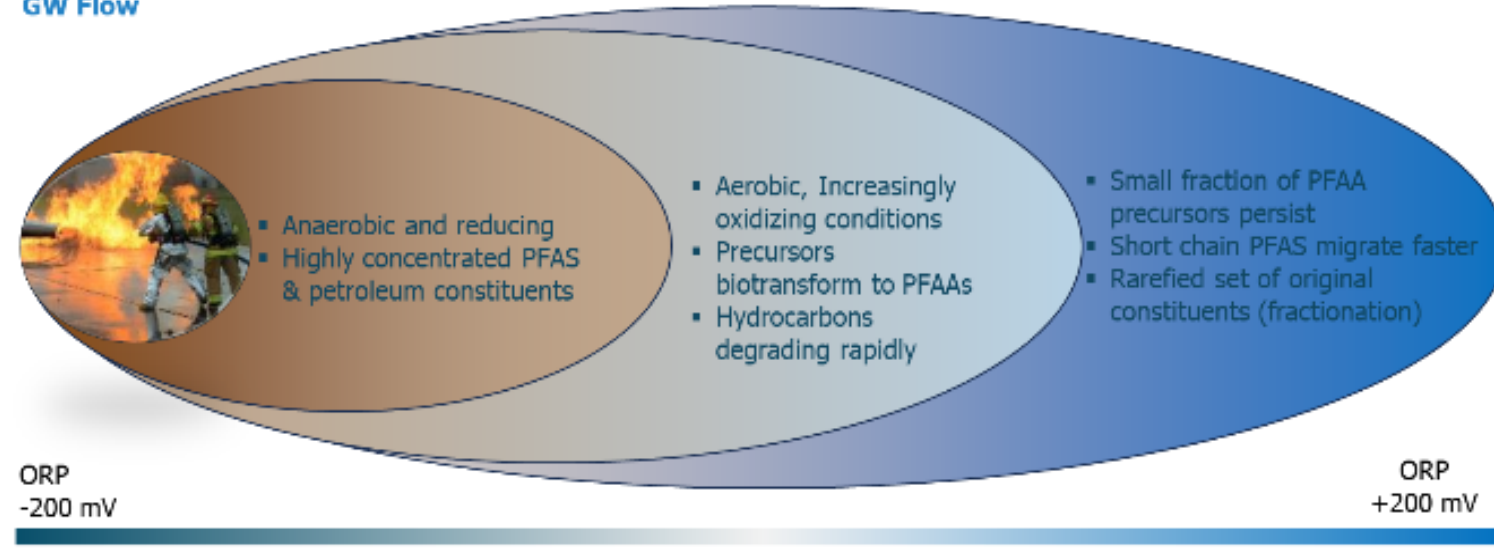
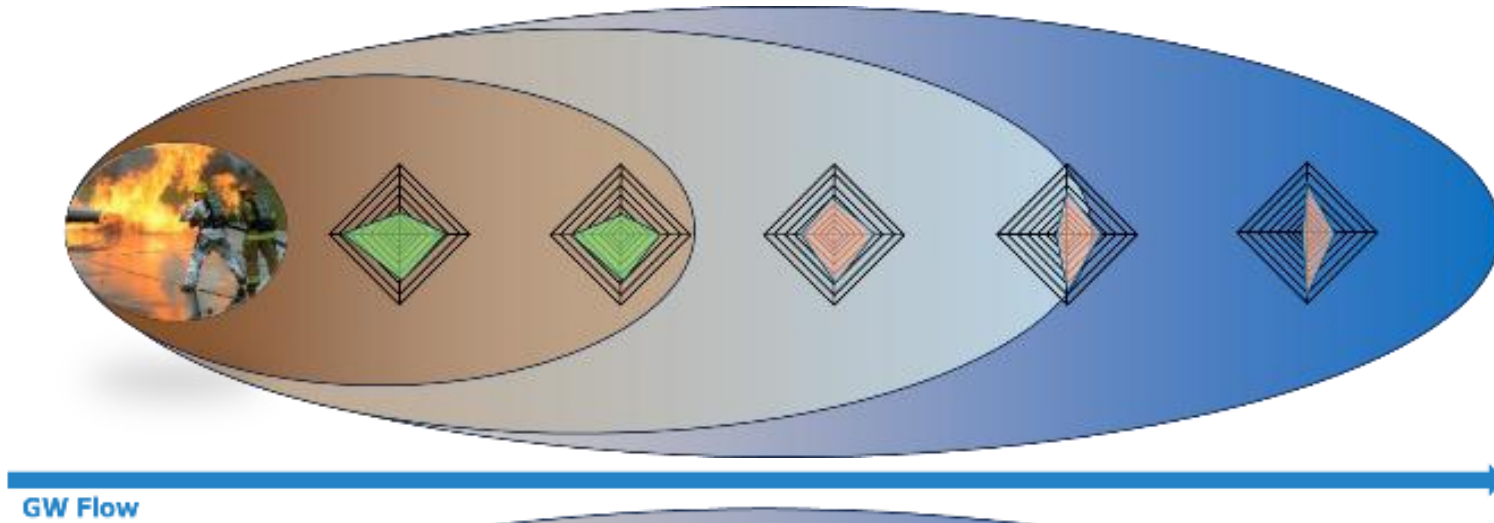
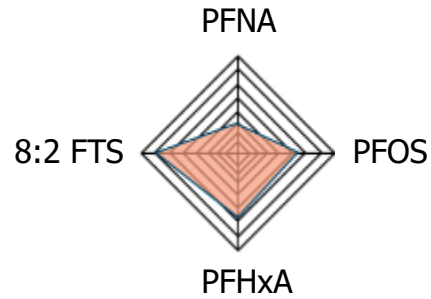


# PFAS Behavior in the Saturated Zone

- Co-contaminants: [Section 1.3.4.1](#)
- Precursor Transformation: [Section 5.4.2](#)
- Diffusion/Back Diffusion: [Section 5.3.1](#)
- Transient Water Level: [Section 5.3.3](#)
- Differential Migration: [Section 5](#); [Section 15.1](#)



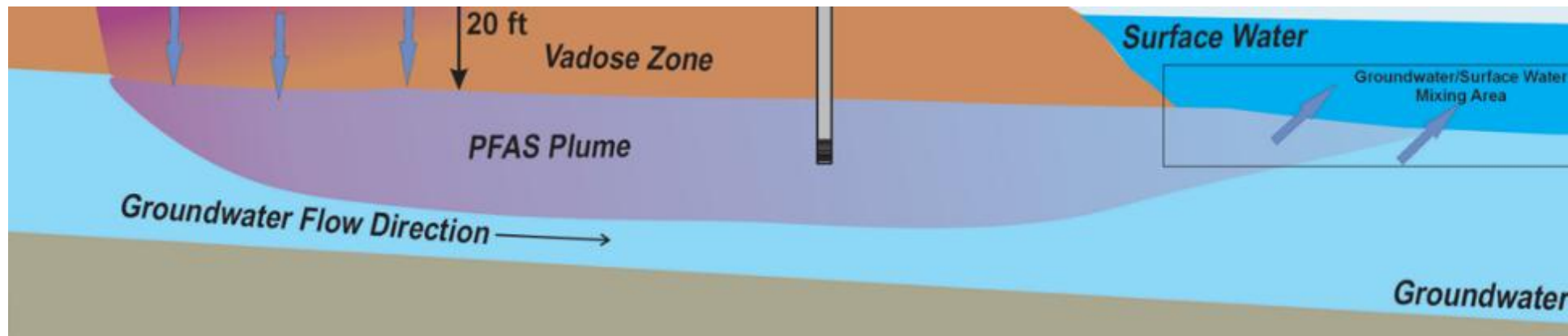
# Differential Transport



A1C Kyle Gese, Public domain, via Wikimedia Commons.

# Sampling Plan for the Saturated Zone

- General Notes For Groundwater Sampling Plan
  - PFAS-free materials, equipment, and supplies: [Section 11.1.7.2](#)
  - Avoid coated bentonite pellets
  - Polytetrafluoroethylene (PTFE) is PFAS
  - Some concrete may contain PFAS
  - PFAS-Free water supply. Confirm water supply meets Data Quality Objectives
  - Minimize turbidity

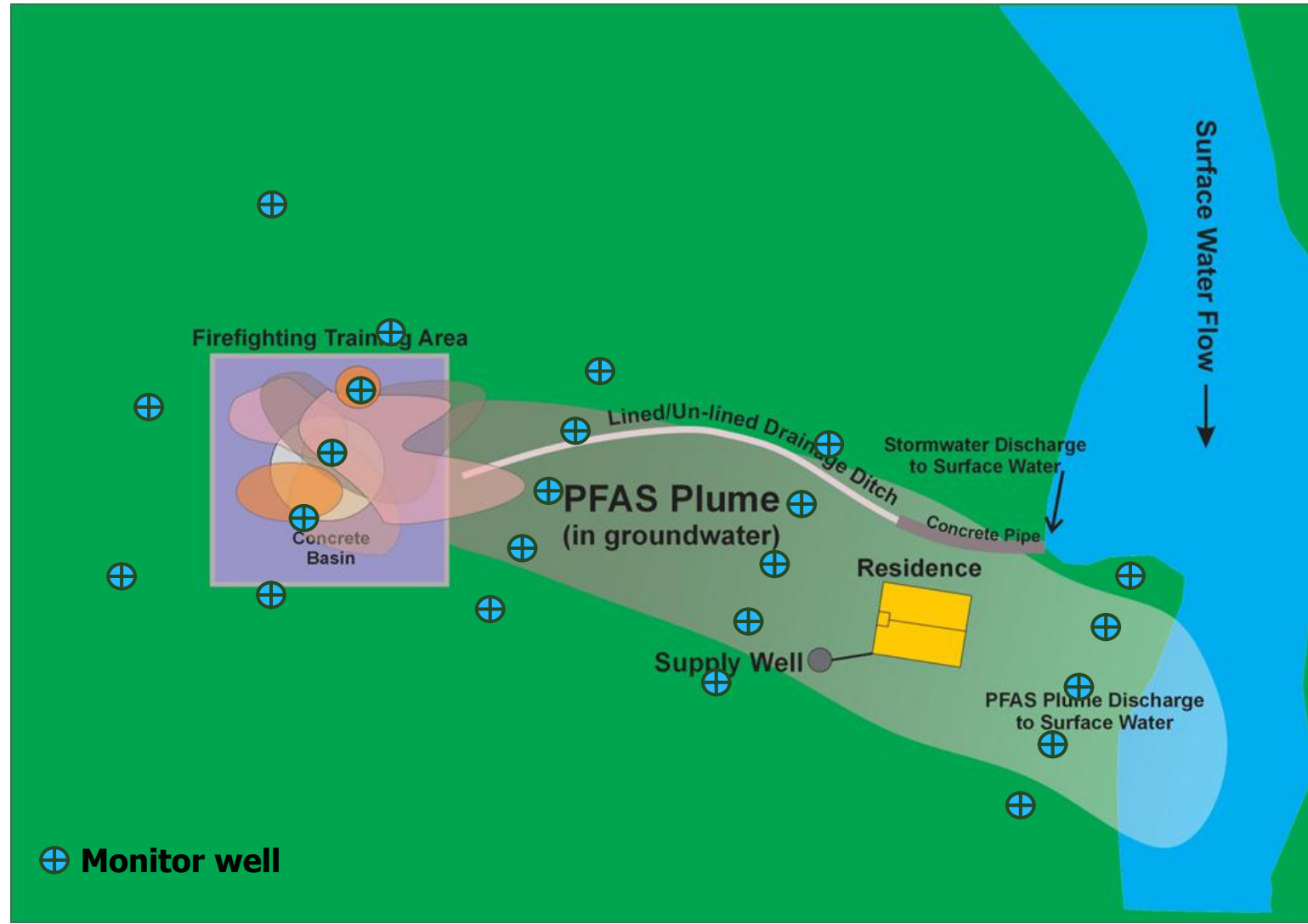


# Sampling Plan for the Saturated Zone

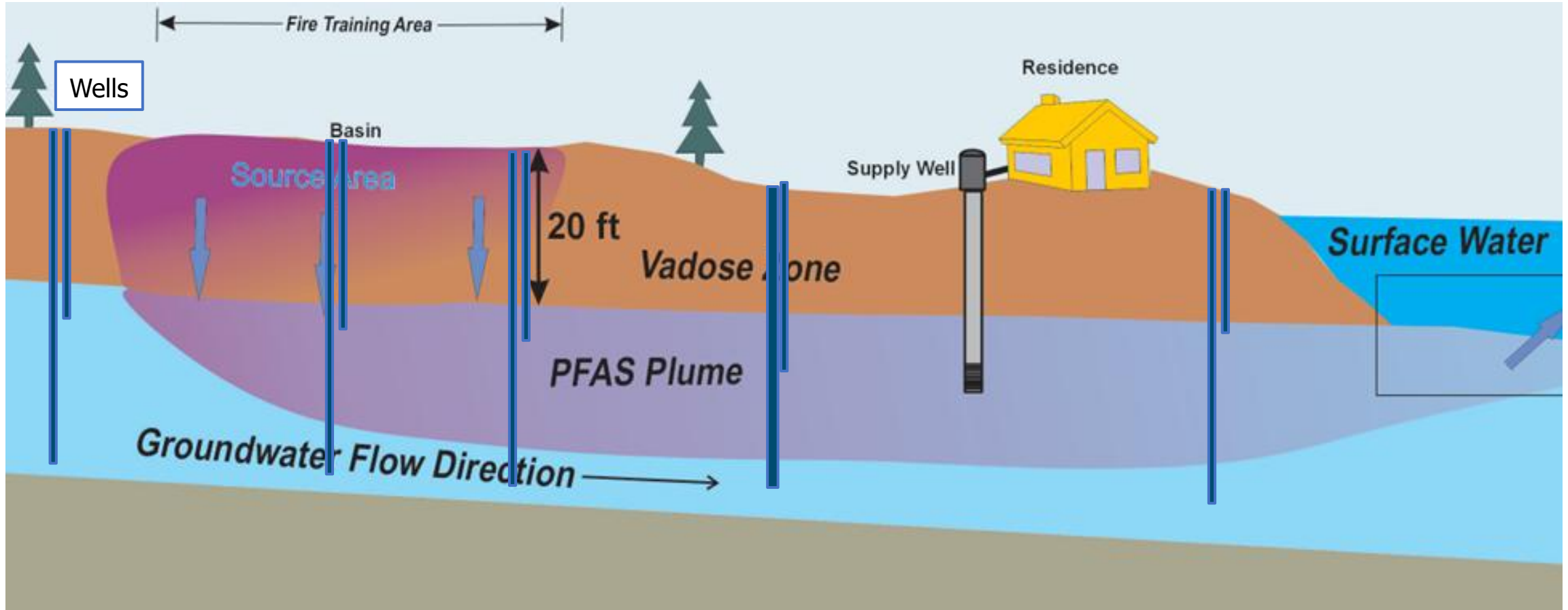
Goal of the saturated zone sampling plan → understand plume characteristics and dynamics

- Fundamentals: Characterize the Hydrogeology (Section 10.3.1.1)
  - Physical, hydraulic, hydrologic, geochemical parameters (may require aquifer testing and geotechnical sampling)
  - Assess site in context of geologic history (e.g., environmental sequence stratigraphy and geomorphology)
- Consider use of High-Resolution Site Characterization (HRSC) tools
  - Direct-Push type tools (MIP, HPT, APS) <https://asct-1.itrcweb.org/>
  - Geophysical tools (surface/borehole/remote sensing)
- Characterize Plume Extent and Stability (Sections 10.4.4 & 10.4.8)
  - Define source area, lateral and vertical extents
  - Consider evaluating mass flux
- Evaluate Precursor Transformation and Impact of Co-contaminants (Sections 5.2.5 & 10.4.6)

# Sampling Plan for the Saturated Zone



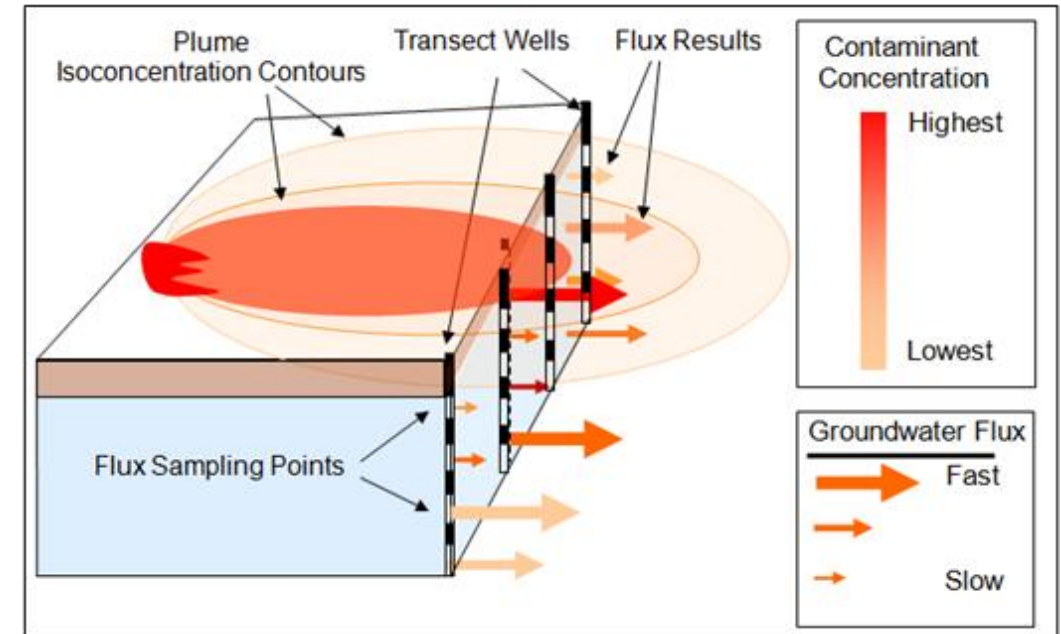
# Sampling Plan for the Saturated Zone



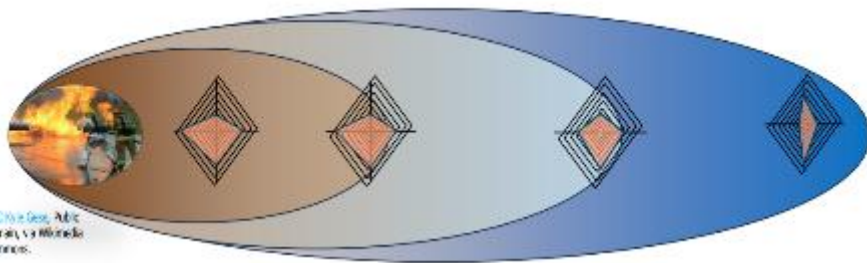
# Data Evaluation for the Saturated Zone

Goals: Understanding which PFAS are migrating, where and how they are migrating, and understanding the time scales involved

- Plume Characteristics
  - Nature and extent, plume stability, mass flux
  - Co-contaminants, seasonality, impacts of extraction
- Tools and Techniques
  - Flow modeling, contaminant fate and transport modeling
  - Forensic evaluation to understand fractionation of constituents
  - Statistical evaluations

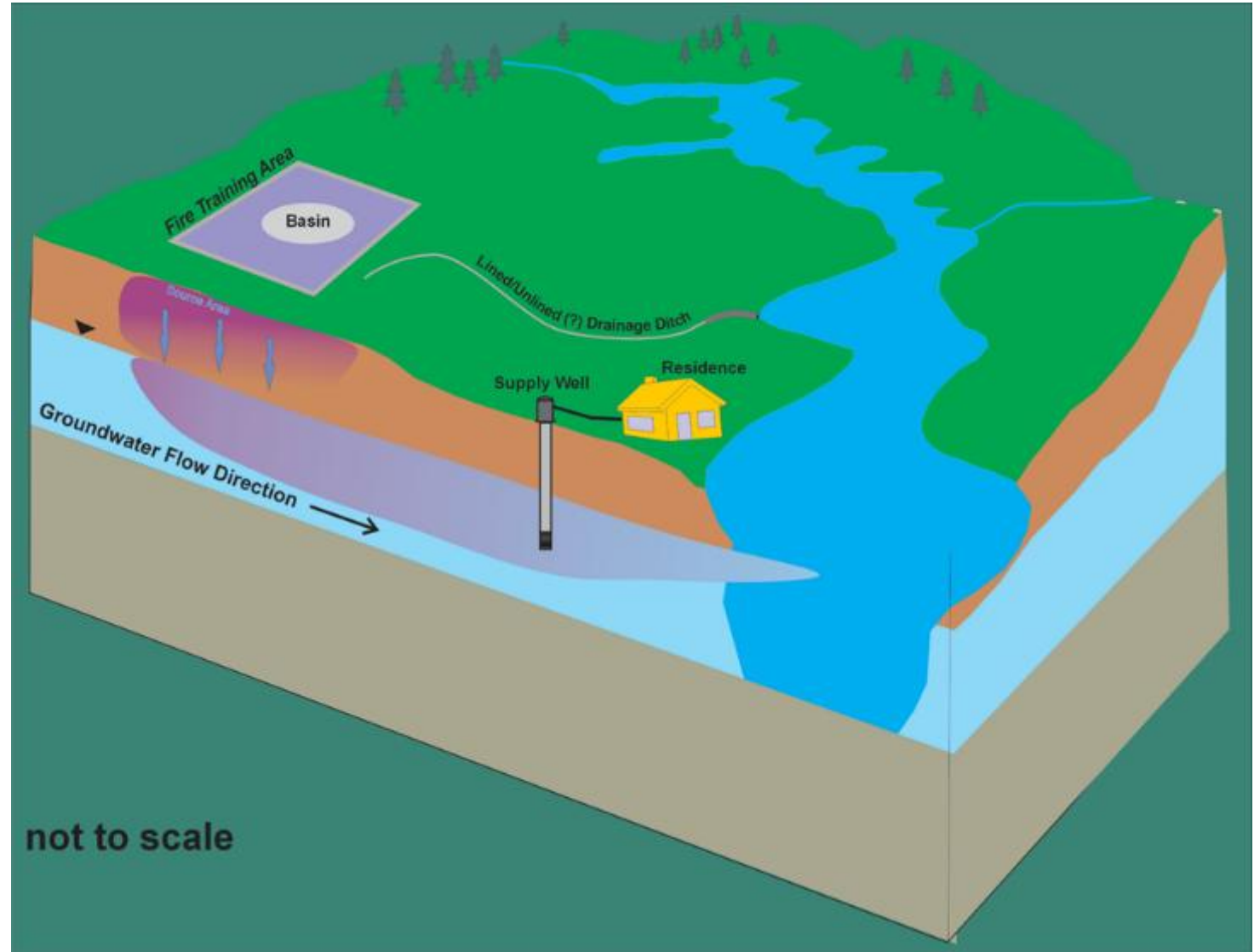


Source: (Graphic courtesy HydroGeoLogic Inc.) Figure 2-4. in ITRC Use and Measurement of Mass Flux and Mass Discharge (ITRC 2010, MASSFLUX-1).



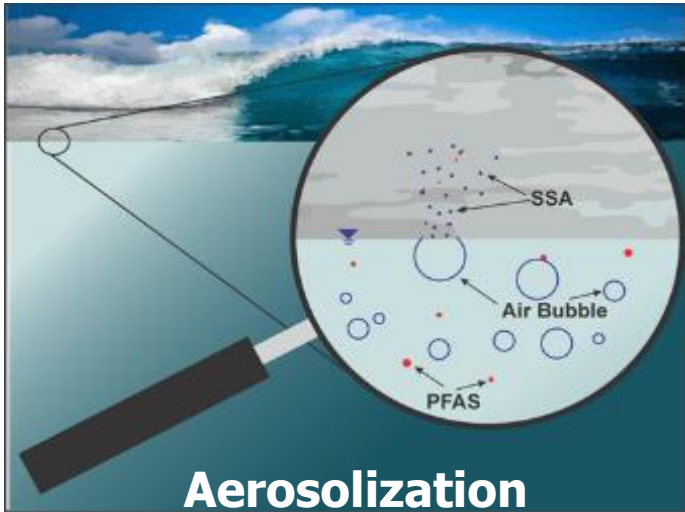
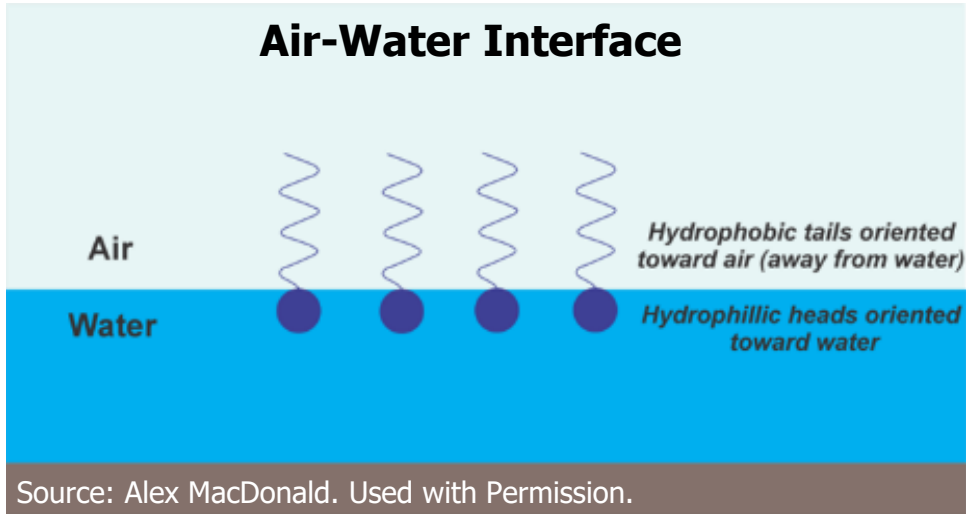
A1C Kyle Gese, Public domain, via Wikimedia Commons.

# Surface Water

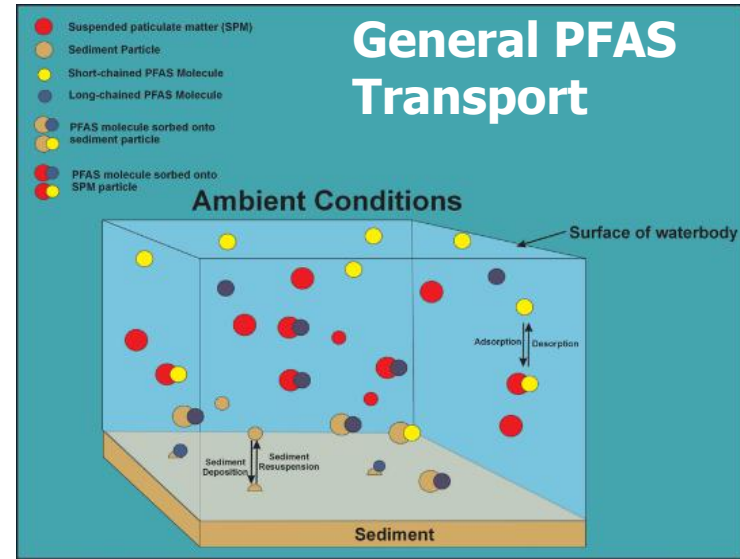


# Surface Water Considerations – Transport in Surface Water

## Air-Water Interface



PFAS-1 Figure 1-4.



PFAS-1 Figure 1-6.

## PFAS-Containing Foam

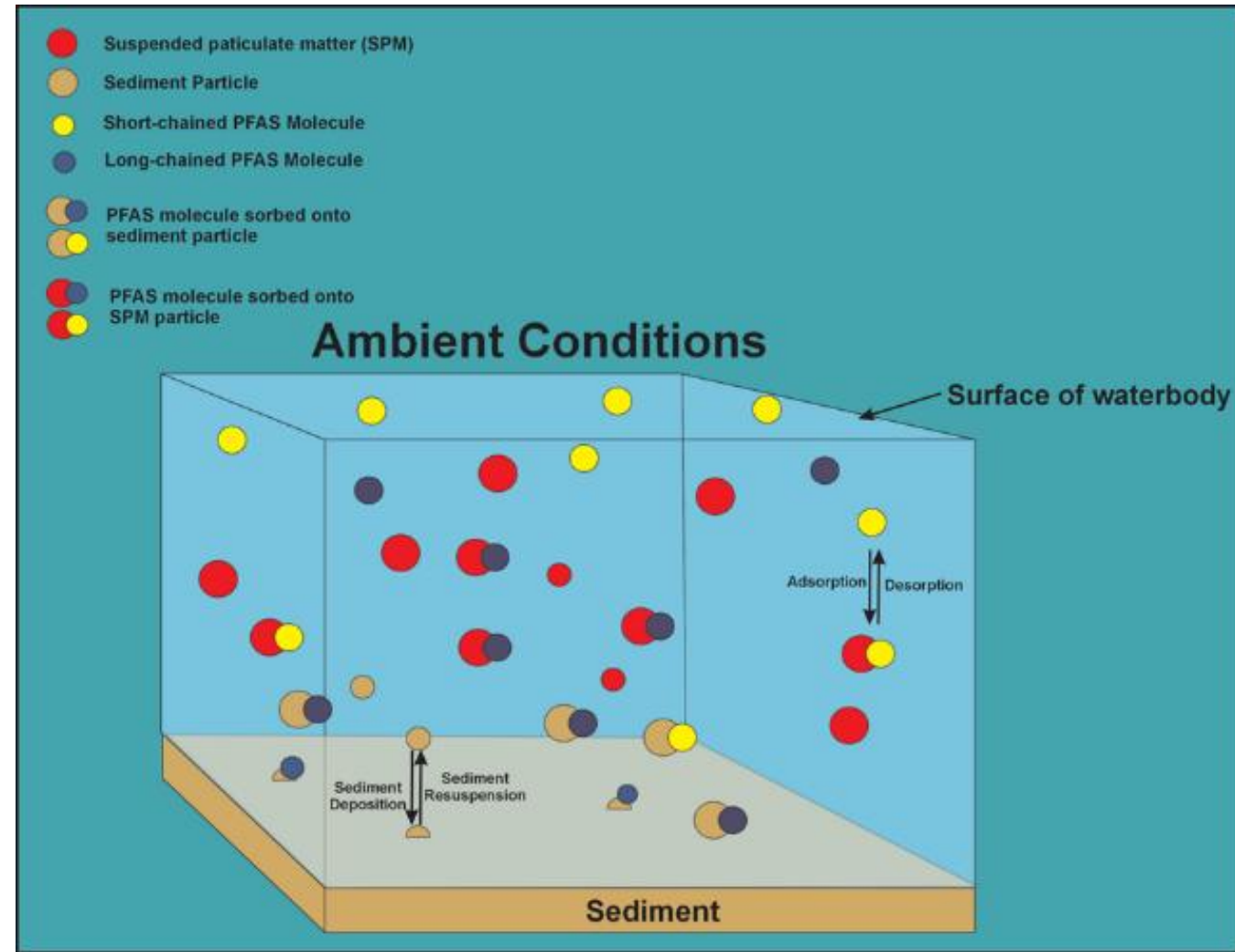
Source: Rebecca Higgins, State of Minnesota. Used with Permission.



# Surface Water Considerations – PFAS Transport in Surface Water

## General PFAS Transport

- PFAS found throughout the water column and sediment – dissolved and sorbed
- More likely short-chained PFAS dissolved and at/near surface
- More likely long-chained PFAS sorbed and/or in sediment and closer to bottom
- Composition/concentration of PFAS sorbed, dissolved, deposited, or suspended continuously changing



PFAS-1 Figure 1-6.

# Surface Water Considerations – Foam

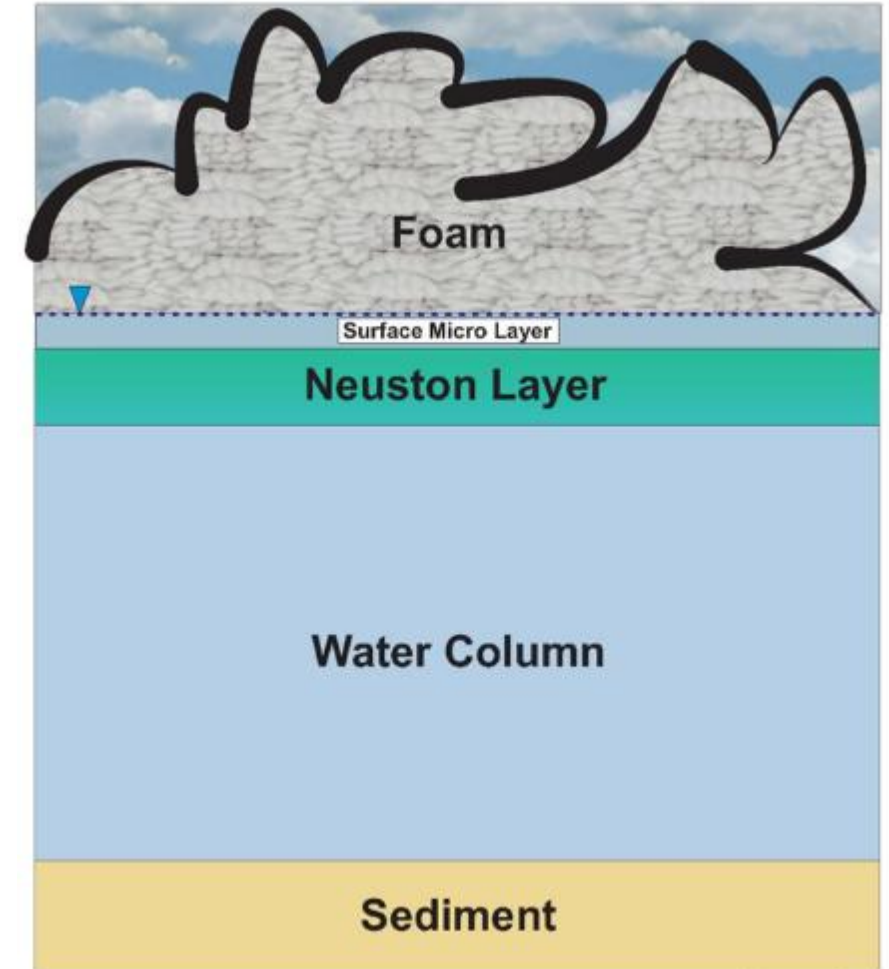
## PFAS-Containing Foam (not AFFF)

- Foam – contains **higher concentrations of PFAS than the underlying layers**; as foam is formed, it removes PFAS from the water column.
- Surface micro layer (about 50 µm thick) - includes the air-water interface. Likely **highest concentration of PFAS in water column**
- Neuston Layer – zone directly below surface micro layer. Rich in aquatic organisms
- Underlying water column

**Transport** as “foam islands” to a new location

Collapse of foam and **dissolution of PFAS** back into water column  
PFAS in foam potentially leads to **additional exposure pathways**  
– both human and ecological receptors

Surface Water Foam PFAS Sampling Guidance – Michigan Dept. of Environment, Great Lakes and Energy July 2019.



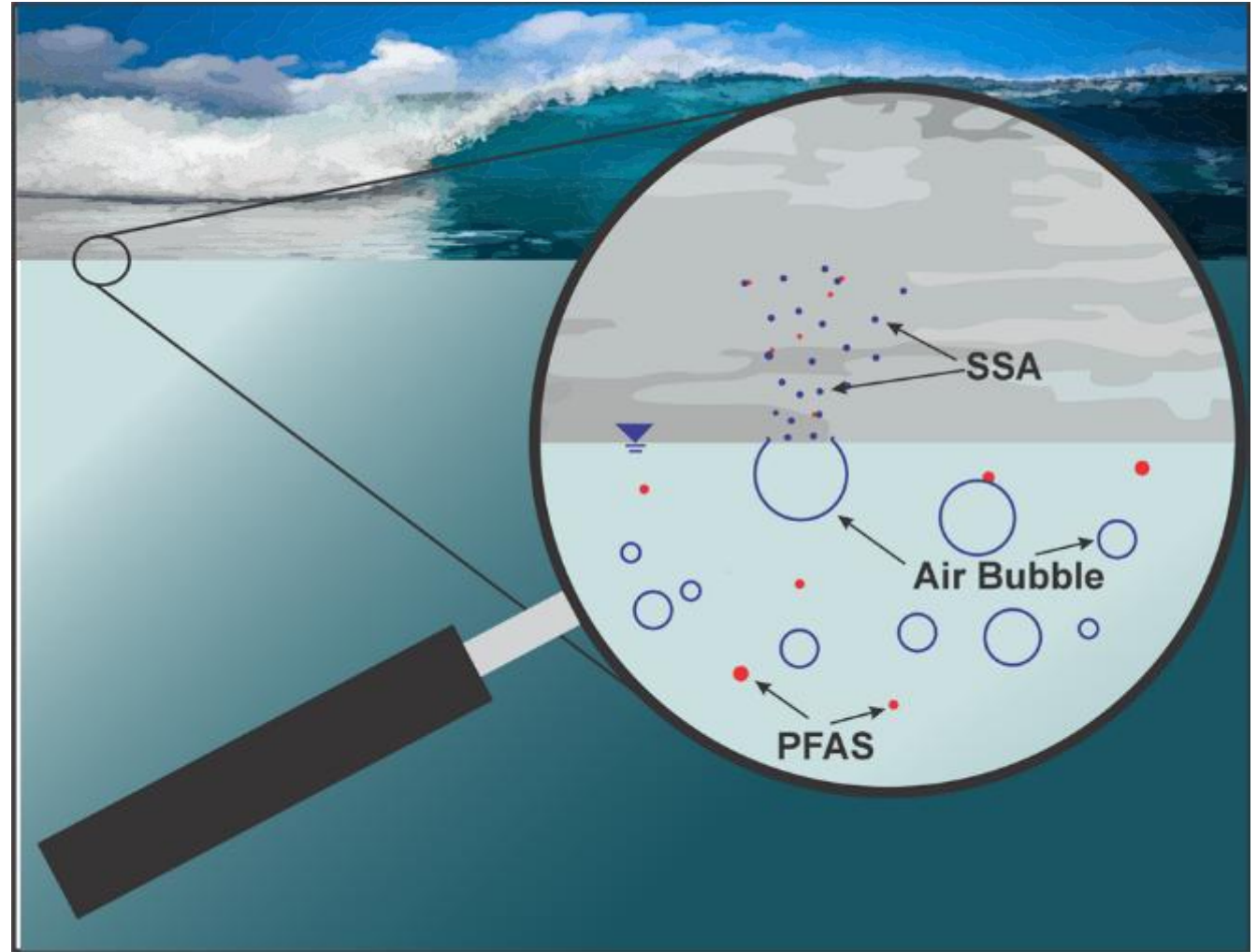
PFAS-1 Figure 16-2.

not to scale

# Surface Water Considerations

## Aerosolization

- Wind/Wave action
- Air bubbles pushed underwater and scavenge PFAS
- Bubbles rise to the surface and burst
- Wind disperses aerosolized particles – sea spray aerosols (SSAs)
- Can be transported hundreds of miles
- Can be largest source of exposure to PFAS
- Additional Information – Section 1.3.5.3, including foam

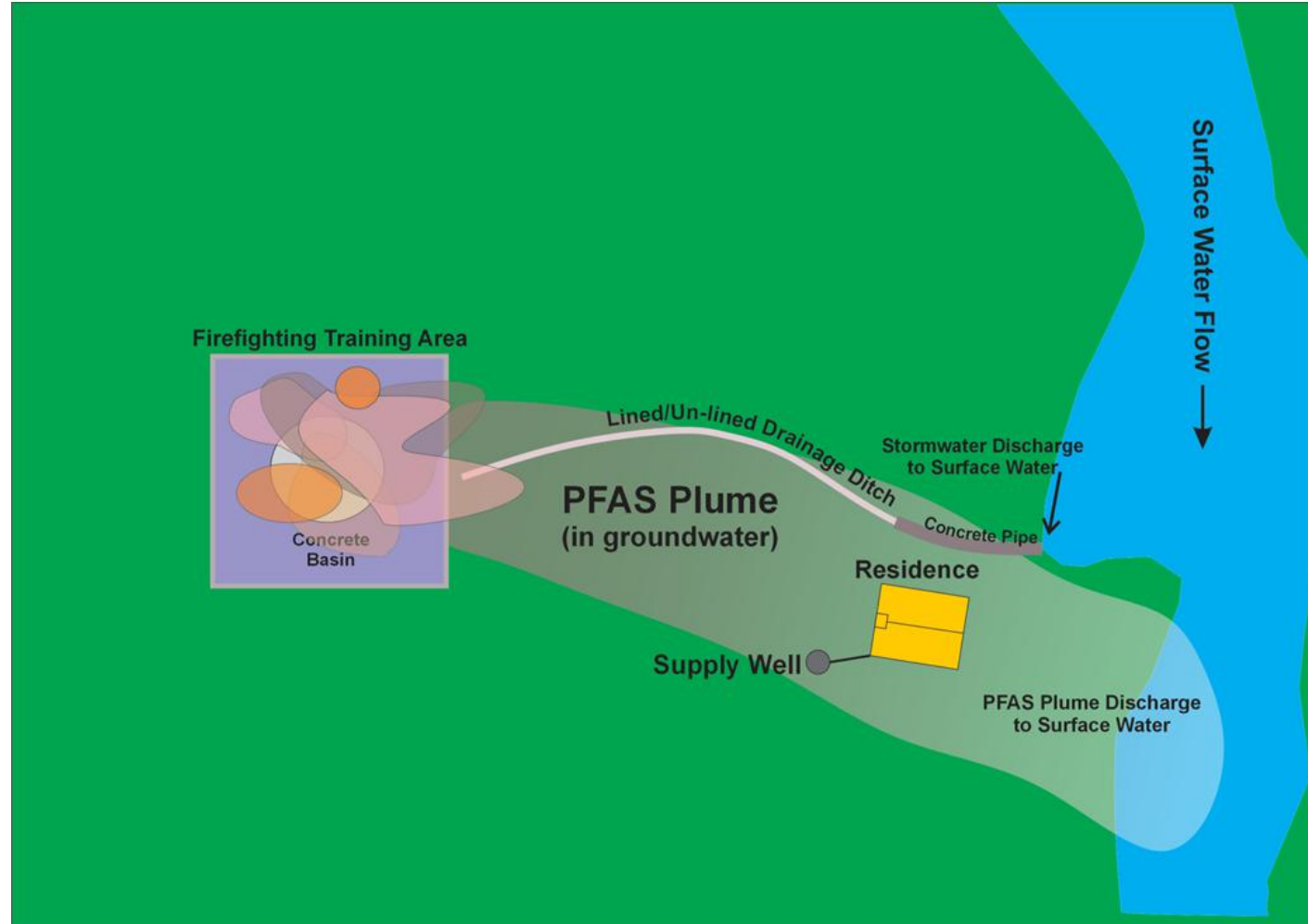


PFAS-1 Figure 1-4.

# Surface Water – Case Study

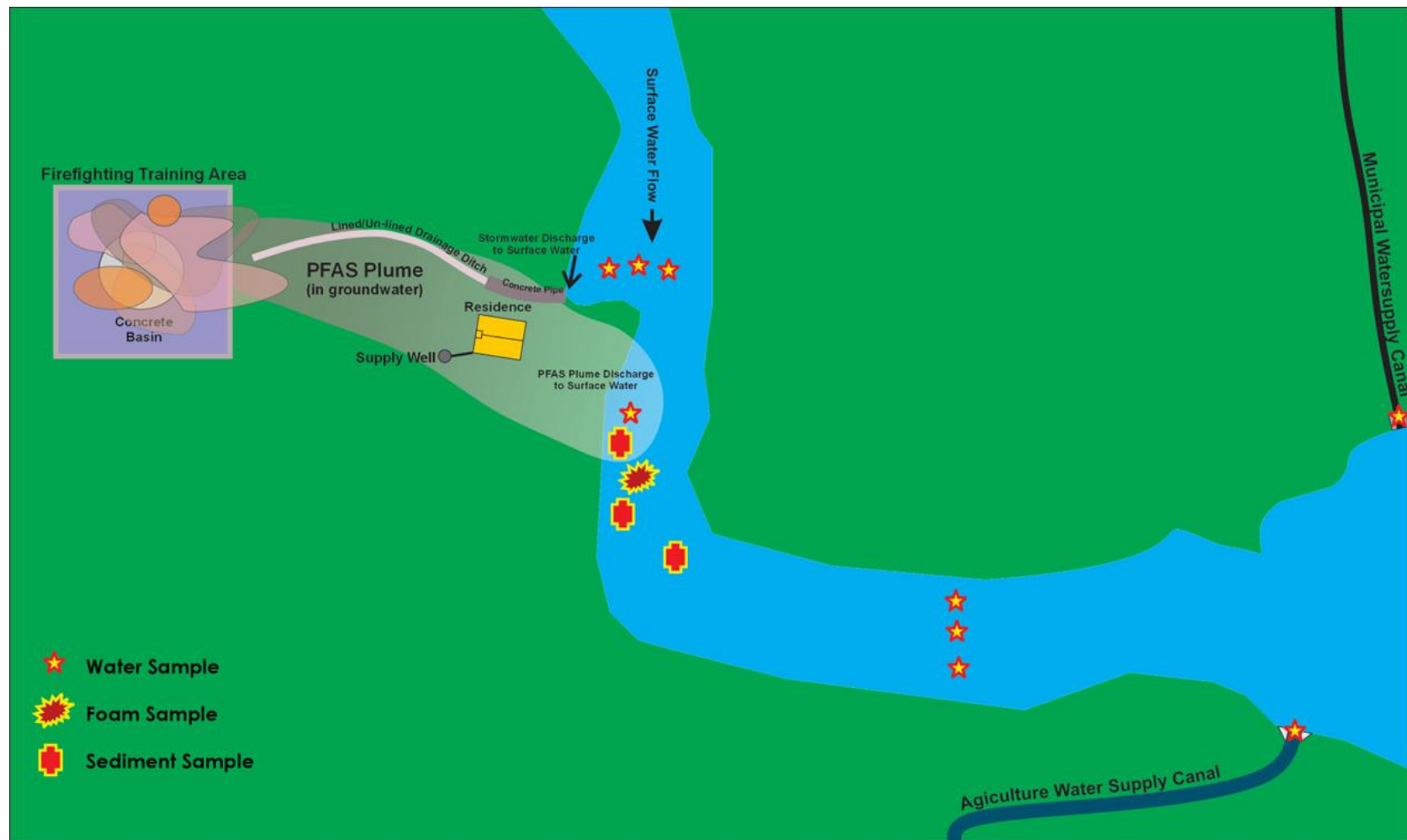
## Overview

- Non-stormwater runoff period
- Impacts are only from plume in groundwater discharging to surface water
- Assess impacts on surface water quality – focusing on protection of human health, aquatic and avian species
- Evaluating Human Health – Sections 16.2 and 7.1
- Evaluating Ecological Risk – Sections 16.3 and 7.2



# Surface Water – Sampling Considerations Case Study

- Low stream flow vs high stream flow
- Upstream and downstream
- Across the water column
- Across the stream
- Occurrence of foam
- Water supply intakes
- Species of concern  
(Section 16.3)



# Knowledge Check!

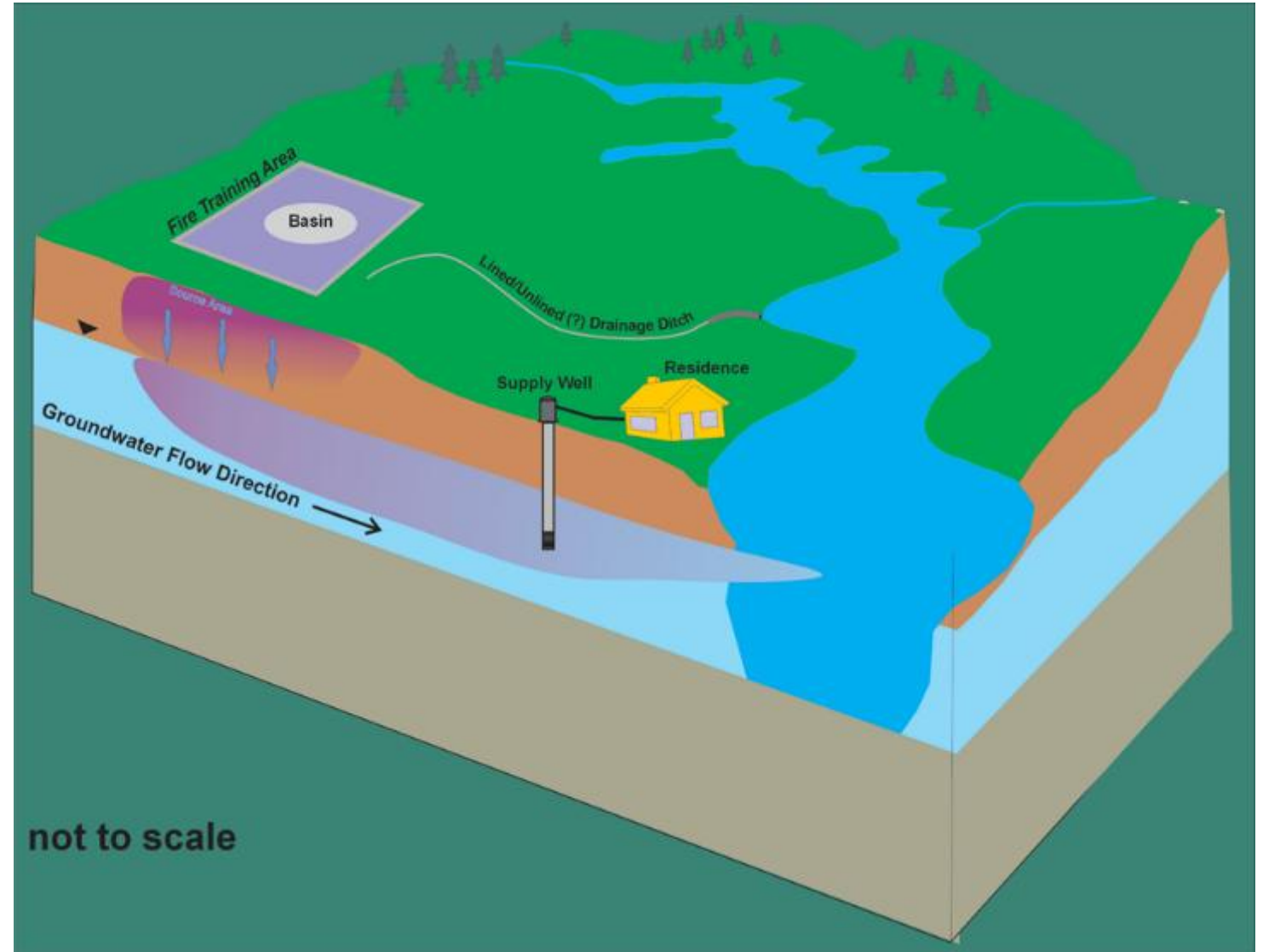
Check  
In!

On a surface water with PFAS-containing foam, where are the highest concentrations of PFAS found?

- A. The surface micro layer
- B. The foam
- C. The Neuston Layer
- D. The deeper water column



# Stormwater



# PFAS Behavior in Stormwater

Considerations for stormwater are similar to those for surface water

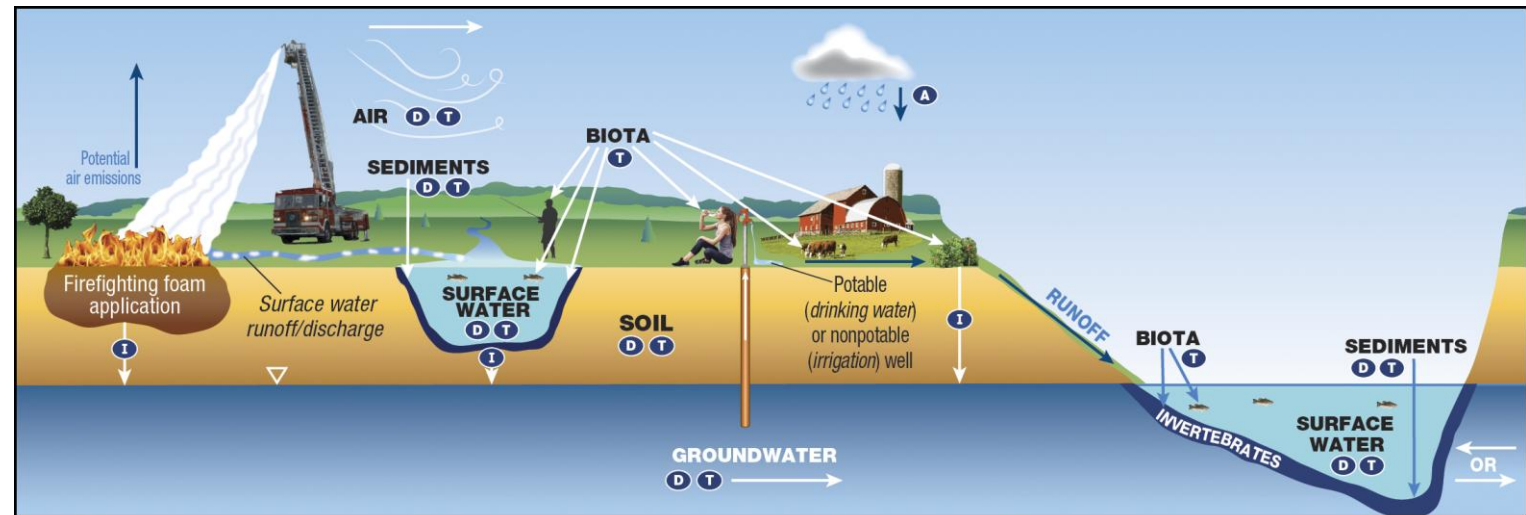
- Foam formation, aerosolization, impacts of interfaces, suspended solids

Aspects to consider

- Inherent variability (intensity, frequency, duration)
- Stormwater runoff
- Impact of runoff on receiving waters

PFAS sources

- Atmospheric
- Site related
- Concrete as a residual source



KEY A Atmospheric Deposition D Diffusion/Dispersion/Advection I Infiltration T Transformation of precursors (abiotic/biotic)

PFAS-1 Figure 2-21. Source: Adapted from figure by L. Trozzolo, TRC. Used with permission.

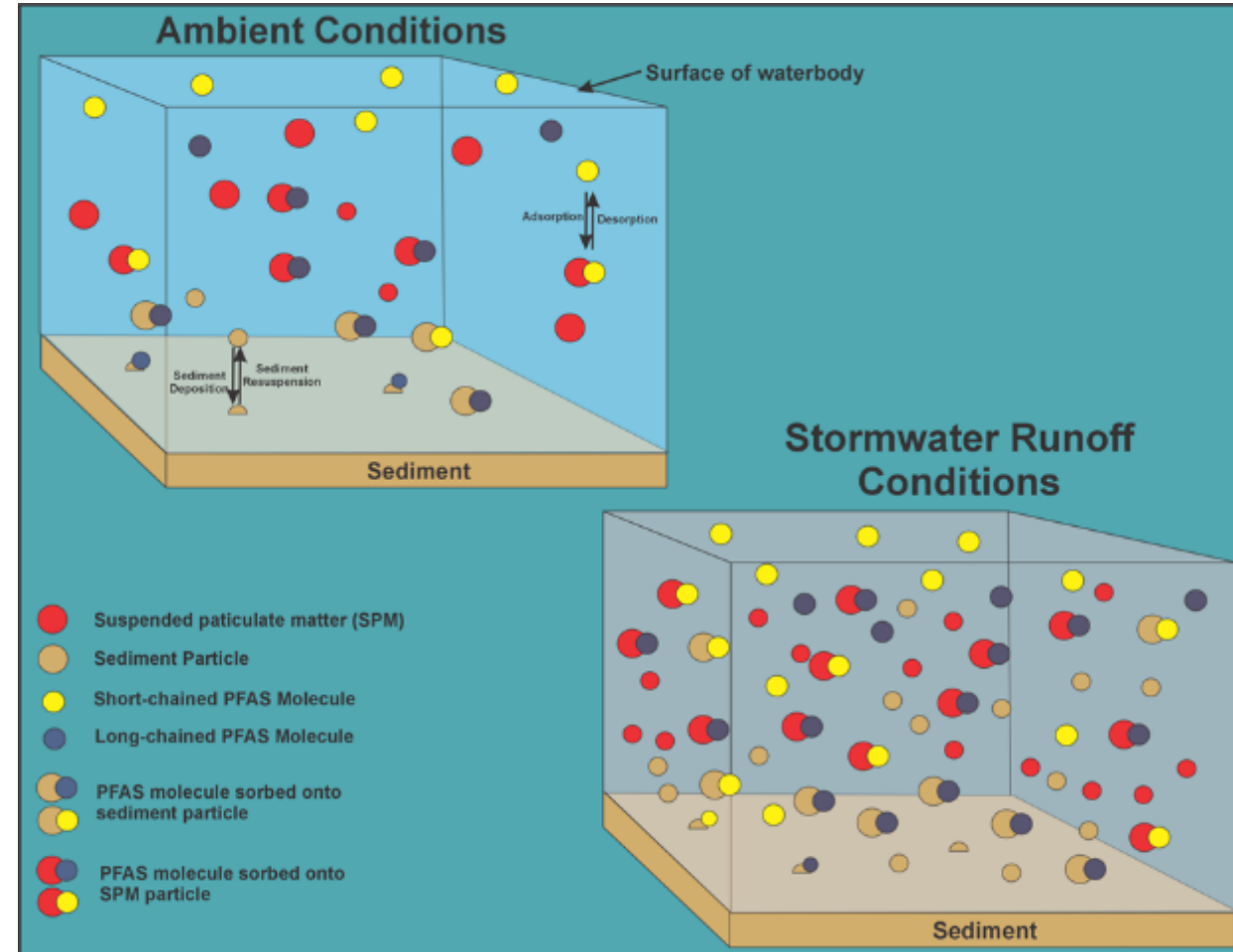
# Sampling Plan for Stormwater – Case Study

## Sampling should address

- Suspended solids and dissolved organic matter provide additional sites for PFAS sorption
- Importance of first flush
- Increased velocity increases transport distance
- Increased turbulence increases mixing

## Additional considerations

- Concentrations of PFAS may be lower, but total mass flux may be greater
- Analysis of solids portion of the water sample
- Foam formation potentially increased
- Foam can act as a source and may become more mobile
- Rain may also be PFAS source to water

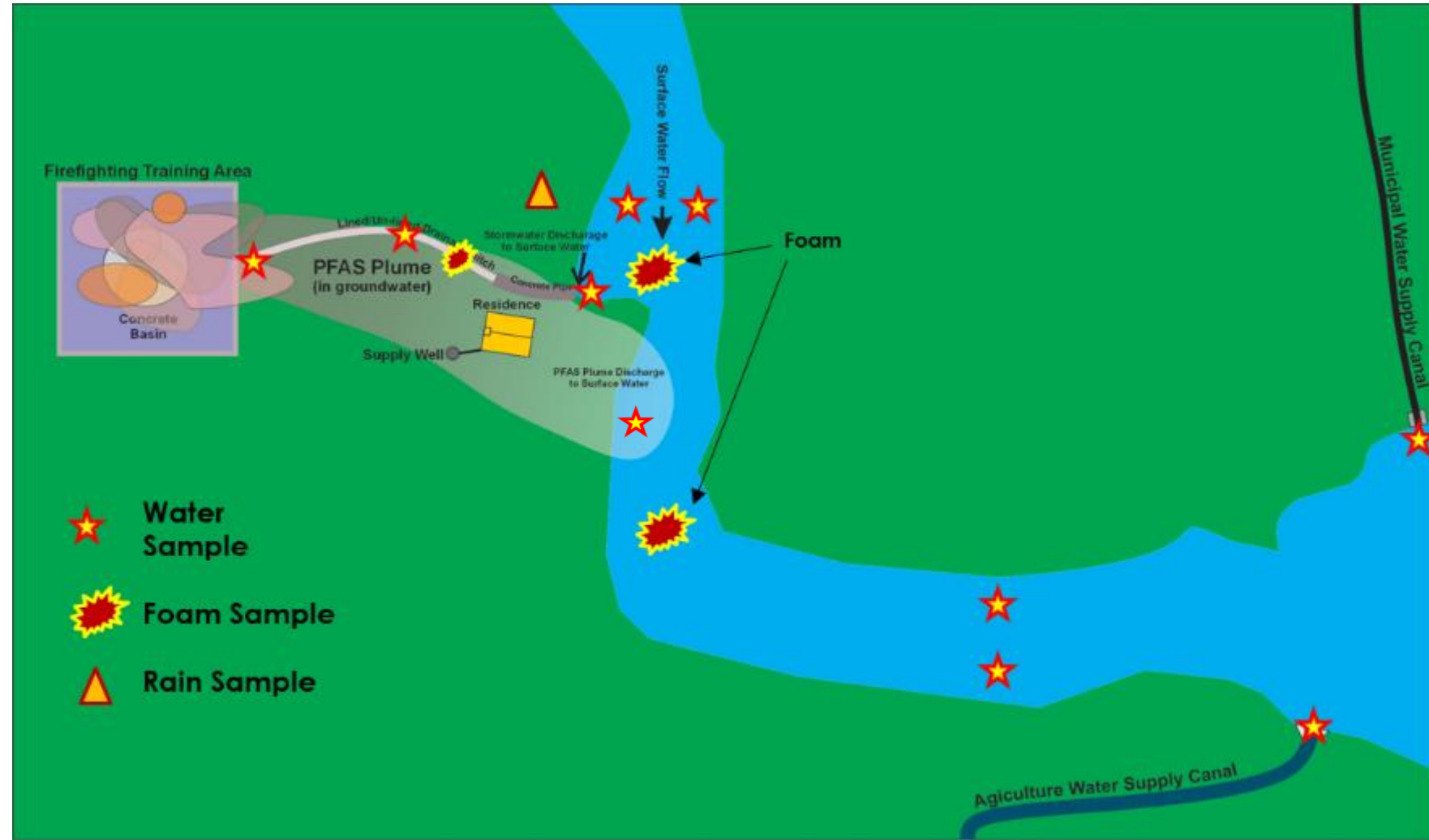


PFAS-1 Figure 1-6.

# Sampling Plan for Stormwater – Case Study

## Sample Locations:

- Upstream (baseline)
- Downstream
- Runoff from Source Area
- Mid-Points
- Discharge at Outfall
- Foam
- Supply Intakes
- Rain Sample

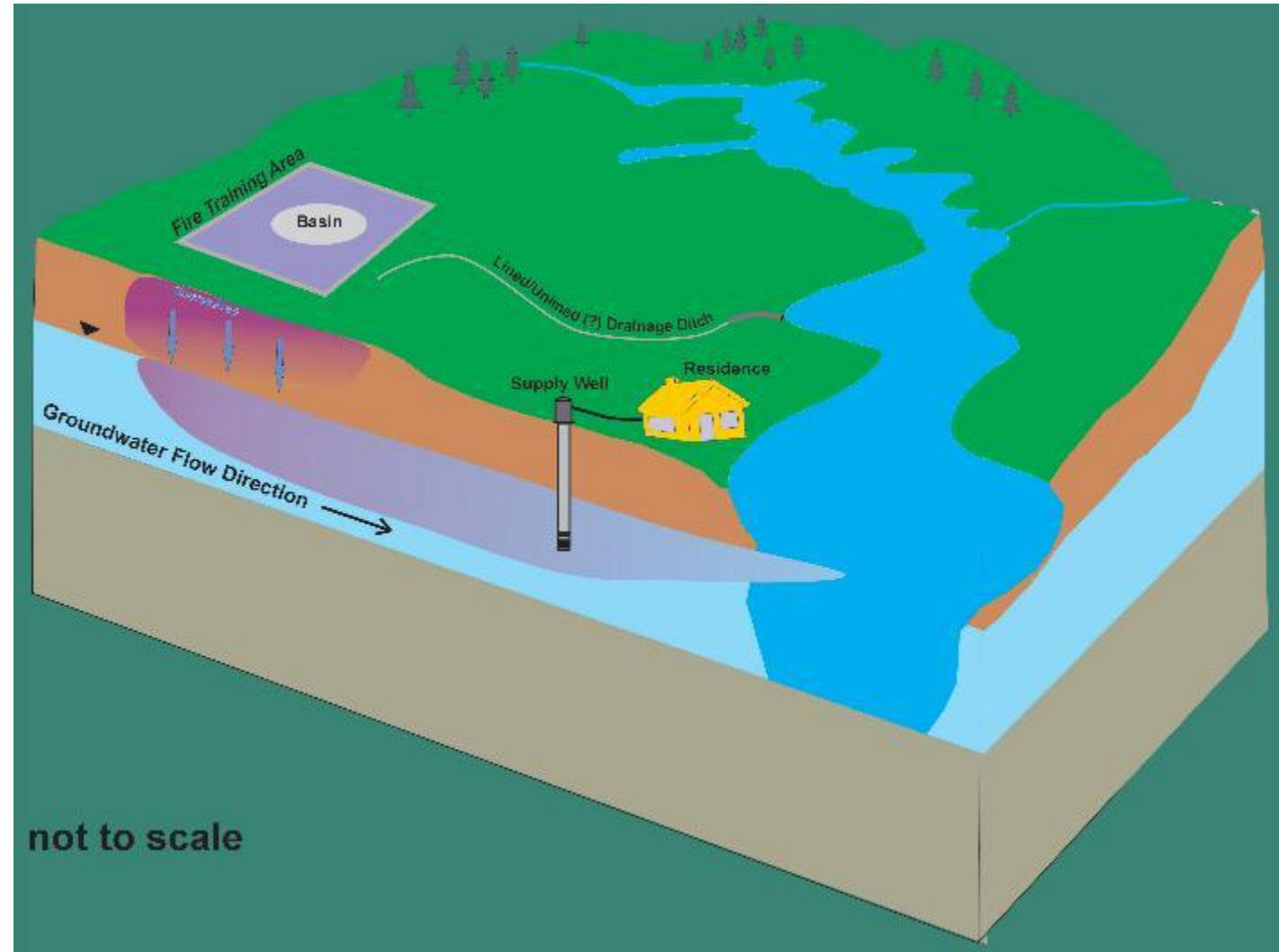


# Data Evaluation for Stormwater

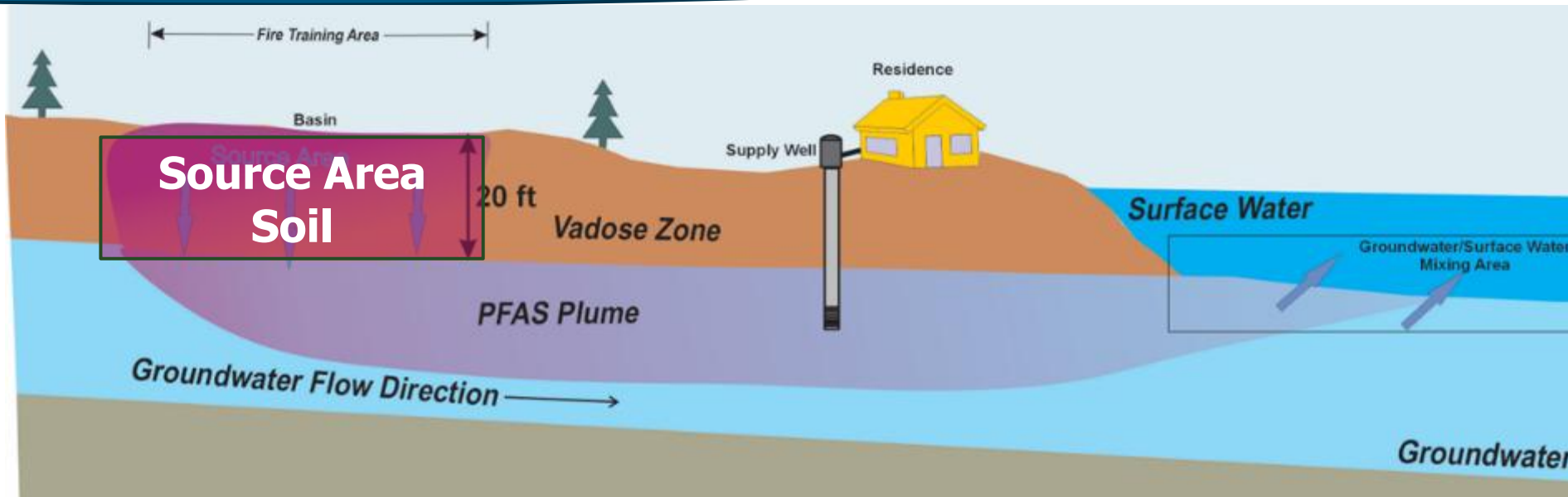
Goals: Understanding how stormwater may/may not transport PFAS and the factors effecting the fate of PFAS in stormwater

- Concentration vs Mass Loading
  - Which is more important to the project objectives?
  - High concentrations (ng/L) at low rates (L/min) compared to low concentrations at high rates.
- Understanding First Flush at Site
  - Concentrations during first flush vs concentrations during latter periods
  - Assess mass migration related to first flush vs remainder of event
- Seasonality
- Potential Residual Source(s)

# Considerations for PFAS Remediation



# Considerations for Soil Remediation



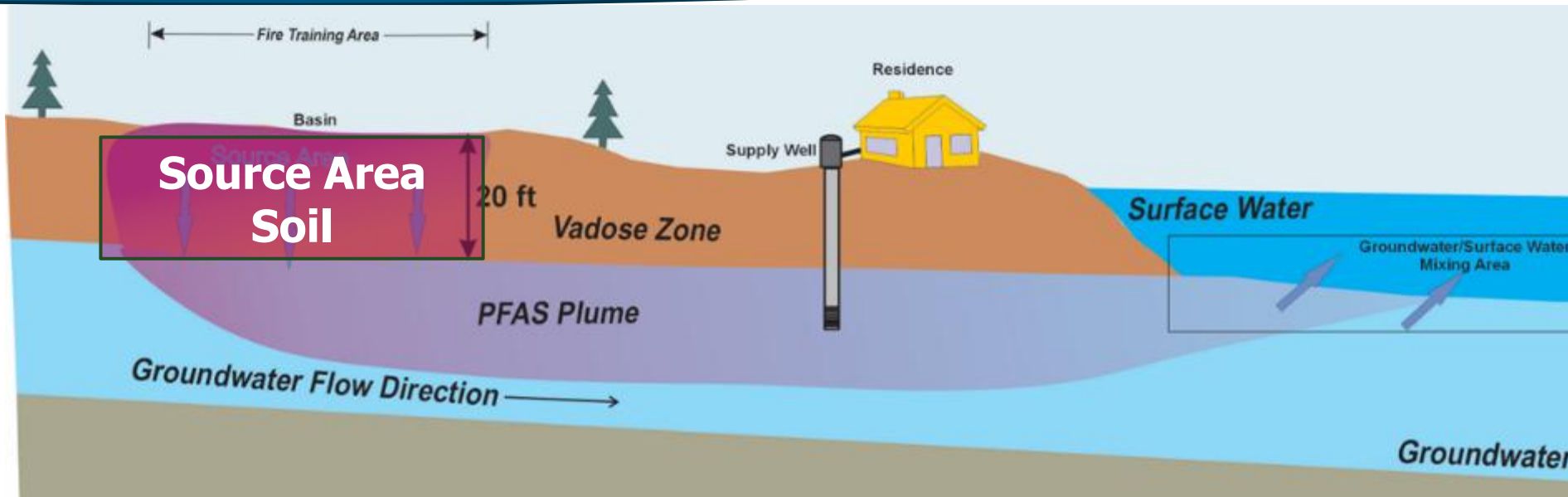
## PFAS Fate & Transport

- PFAS source strength (total PFAS, precursors)
- PFAS composition (regulated/non-regulated, precursors)
- Partitioning, retention & phase distribution

## PFAS Remediation – Soil

- Remedy selection, potential impact of remedy on unregulated PFAS, remedy resilience
- Mass balance and products of incomplete destruction (destruction technologies, e.g., incineration/thermal)
- Selection and dosage of amendment (sorption/stabilization) or reagent (soil washing)

# Considerations for Soil Remediation (cont'd)



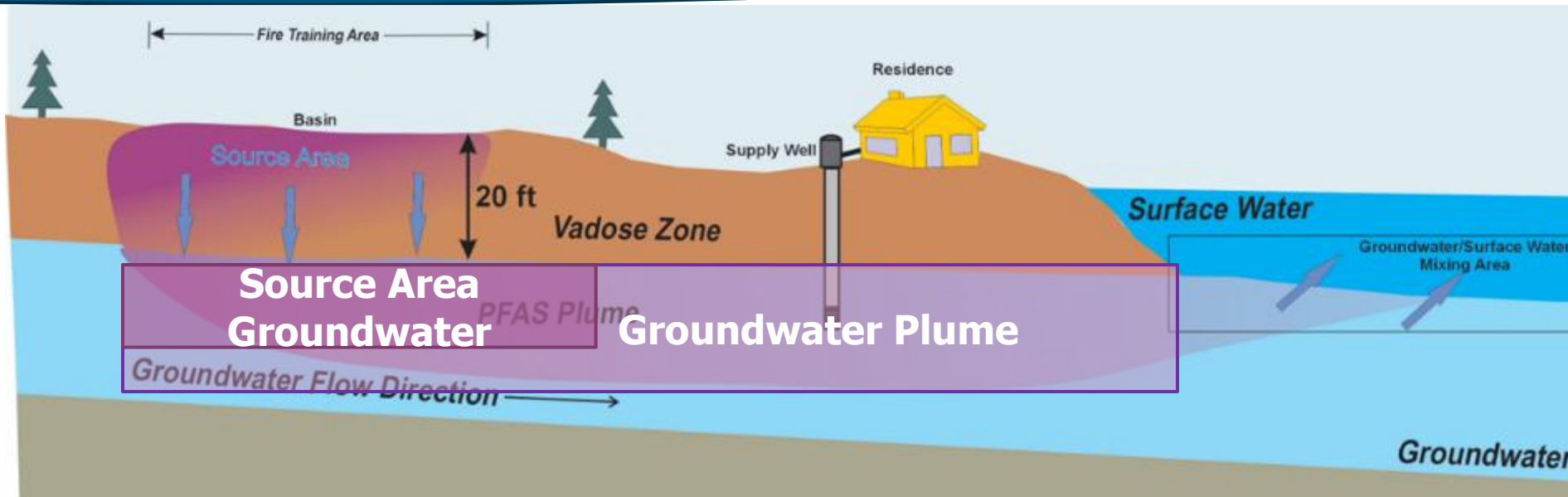
## PFAS Fate & Transport

- Interaction with co-contaminants
- Precursor transformation
- Leaching potential

## PFAS Remediation – Soil

- Long-term stability/performance (sorption/stabilization)
- Final disposal considerations (e.g., landfill)

# Considerations for Groundwater Remediation



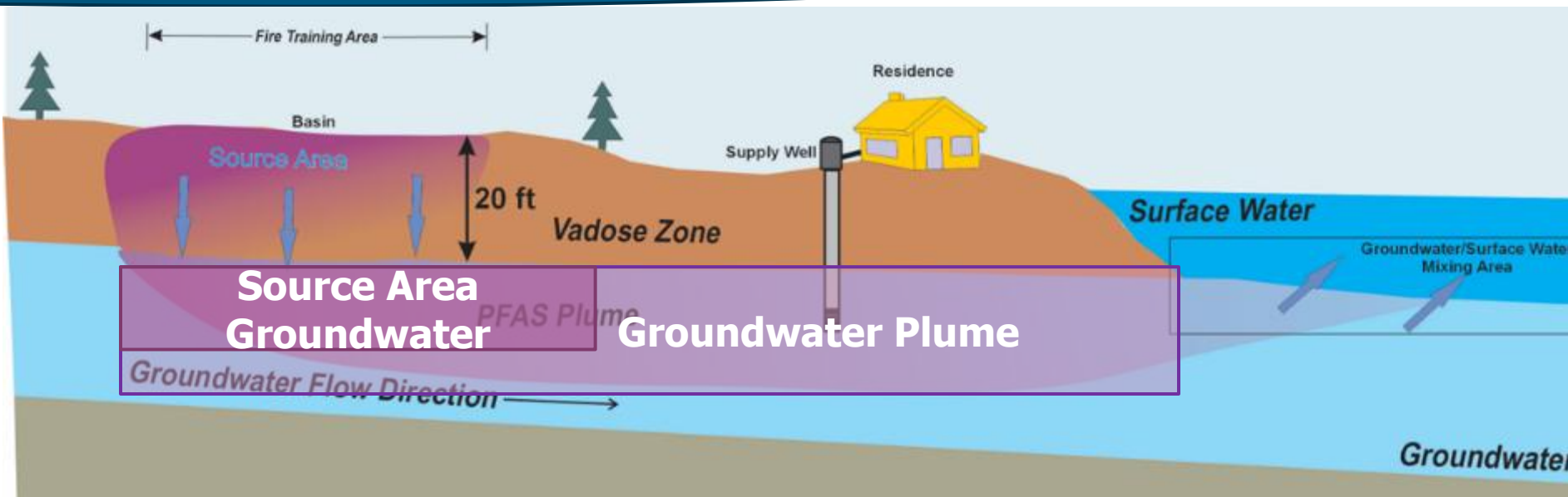
## PFAS Fate & Transport

- PFAS composition (regulated/nonregulated, precursors)
- Co-contaminants
- PFAS source strength (total PFAS, precursors)
- Vadose zone leaching (composition/evolution, magnitude, duration, etc.)
- Precursor transformation

## PFAS Remediation – Groundwater

- Remedy selection, potential impact of remedy on unregulated PFAS, remedy resilience
- Sorption media selection, pre-treatment, system design (ex-situ treatment, e.g., GAC, IX)
- Source & plume longevity (in-situ sorption, MNA evaluation, long-term plume management)

# Considerations for Groundwater Remediation (cont'd)




## PFAS Fate & Transport

- Partitioning, retention & phase distribution
- Diffusion/back diffusion
- Colloidal transport

## PFAS Remediation – Groundwater

- Feasibility of Pump & Treat
- Plume evolution, migration & longevity, mass flux, “clean-up” timeframe (in-situ sorption, MNA evaluation, long-term plume management)

- PFAS Introductory Training 
- PFAS Chemistry Explained
- PFAS & Biosolids: Sources, Occurrence, Transport, and Treatment
- PFAS - Practical Approaches for PFAS Fate & Transport Evaluation
- PFAS Sorption Based Technologies for Separation & Concentration of PFAS from Water

## PFAS Trainings



# Questions

Please use the Q&A Pod to ask questions.



Feedback Form: <https://www.clu-in.org/conf/itrc/PFAS-fate-transport/>